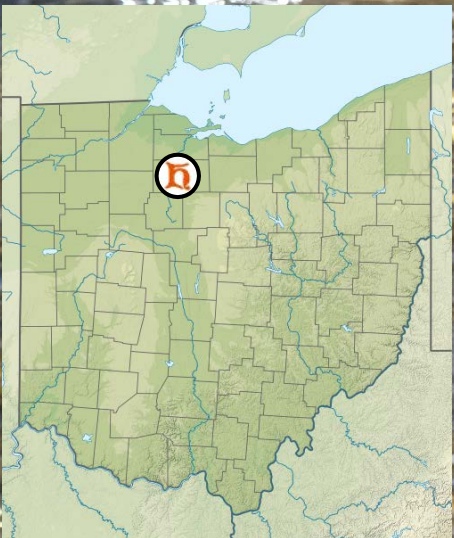


# The Effect of Agriculture on Harmful Algal Blooms

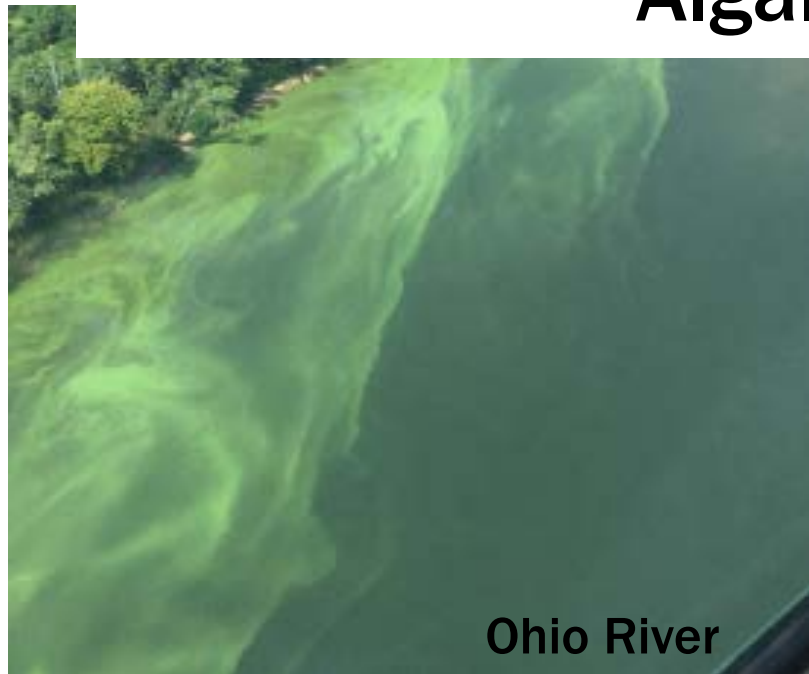
**Laura Johnson**

Director  
National Center for Water Quality Research  
Heidelberg University

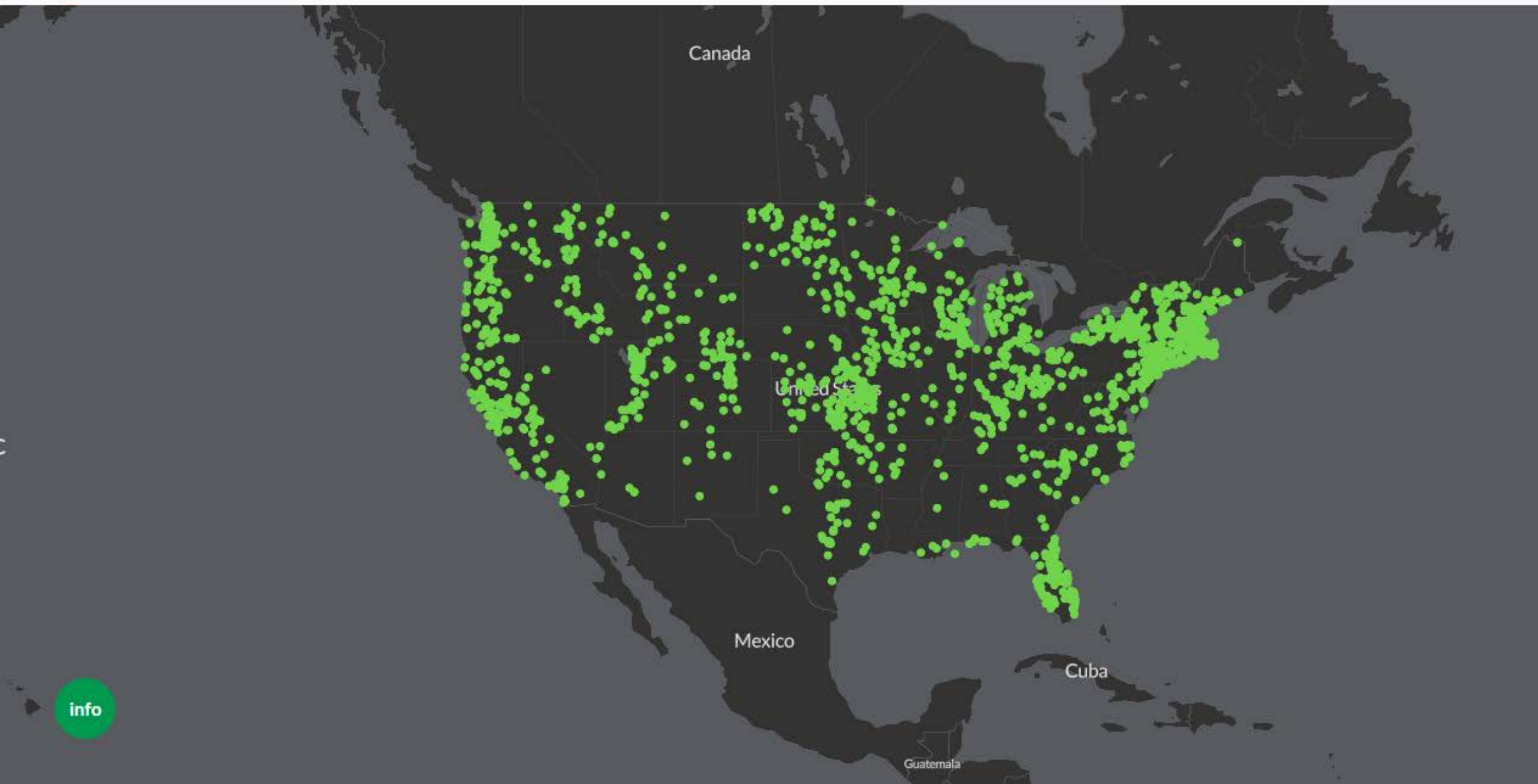




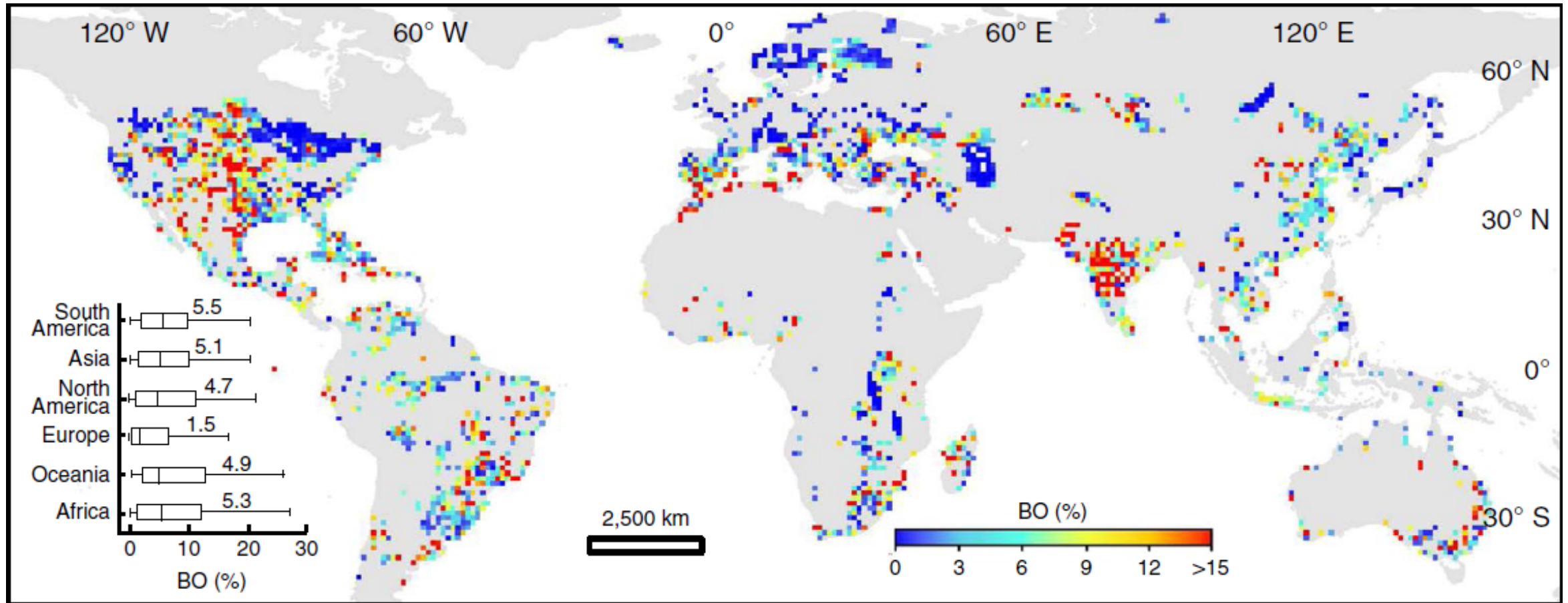
**Algal blooms are common in Ohio**



## News Reports of Algae Blooms, 2010 to Present



# Algal blooms are common globally and are intensifying



Global patterns of lacustrine algal blooms between 1982 and 2019

(BO; that is, the frequency of which algal blooms were detected)

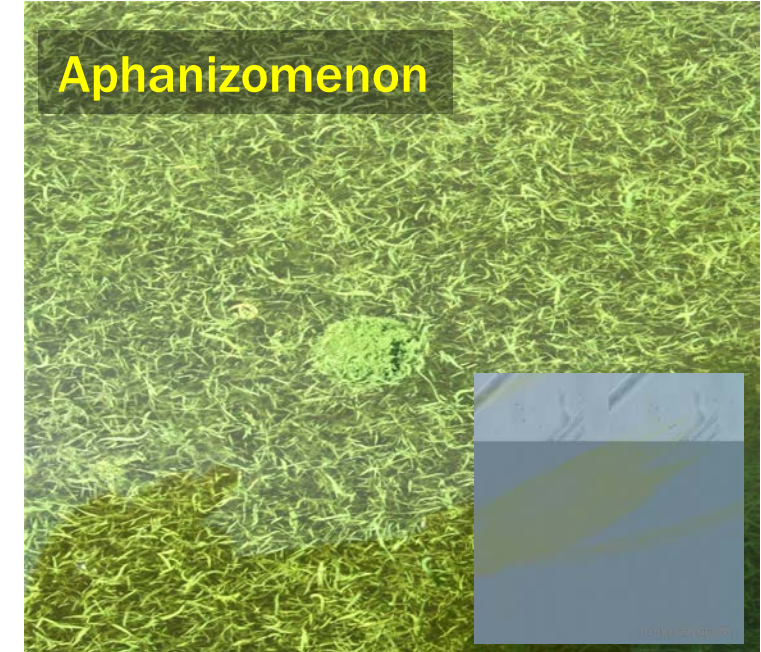
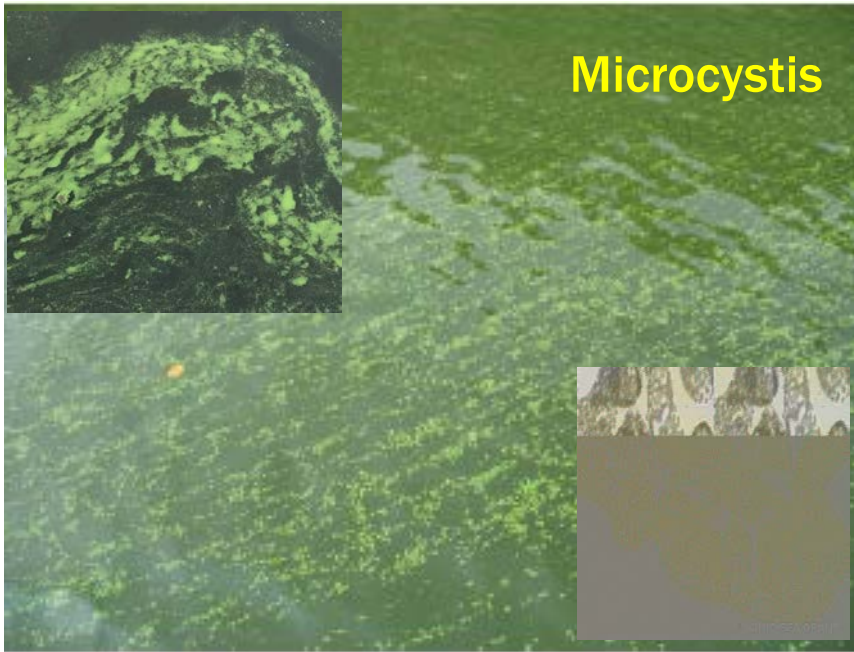
# What are harmful algal blooms?

*\*produce a toxin*

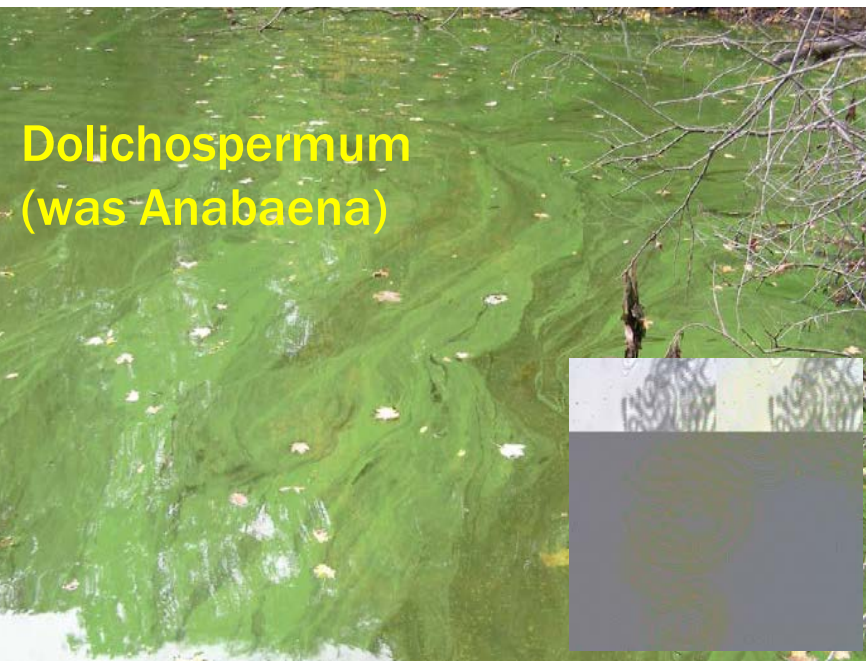
*\*often cyanobacteria*



Lake Erie, 2019  
Aerial Associates Photography



## Toxin-producing cyanobacteria common to Ohio



# Cyanobacterial Toxins



## ANATOXIN

- ☼ Toxin type: *Neurotoxin*
- ☼ Exposure Symptoms: *Staggering, gasping, convulsions, diarrhea, death*
- ☼ Algal Producers: *Anabaena, Aphanizomenon, Planktothrix*



## CYLINDROSPERMOPSIN

- ☼ Toxin Type: *Hepatotoxin*
- ☼ Exposure Symptoms: *Vomiting, diarrhea, liver tumors, death*
- ☼ Algal Producers: *Aphanizomenon, Cylindrospermopsis, Planktothrix*



## SAXITOXIN

- ☼ Toxin Type: *Neurotoxin*
- ☼ Exposure Symptoms: *Numbness, gasping, paralysis, incoherence, death*
- ☼ Algal Producers: *Anabaena, Cylindrospermopsis*



## MICROCYSTIN

- ☼ Toxin Type: *Hepatotoxin*
- ☼ Exposure Symptoms: *Rashes, vomiting, diarrhea, headaches, chest pain, fever, liver tumors, death*
- ☼ Algal Producers: *Anabaena, Microcystis, Nostoc, Planktothrix*



# Why do blooms require to form?

- Light
- Warm temperatures
- Nutrients (nitrogen and phosphorus)

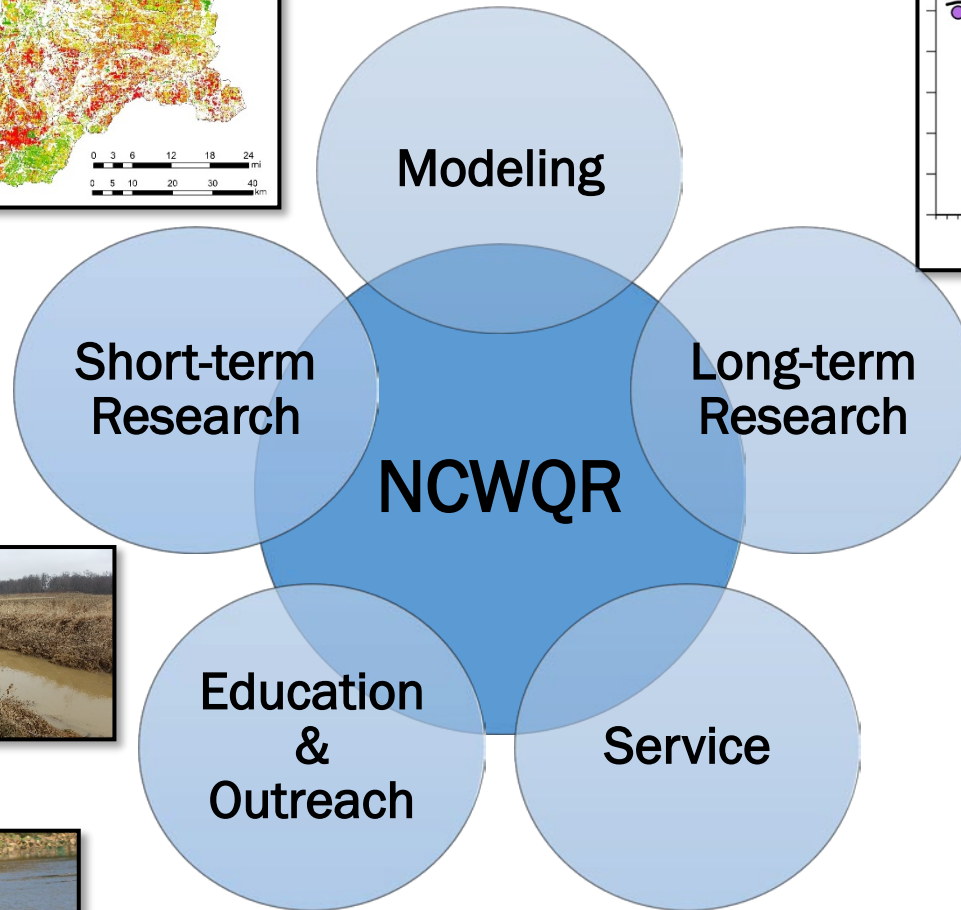
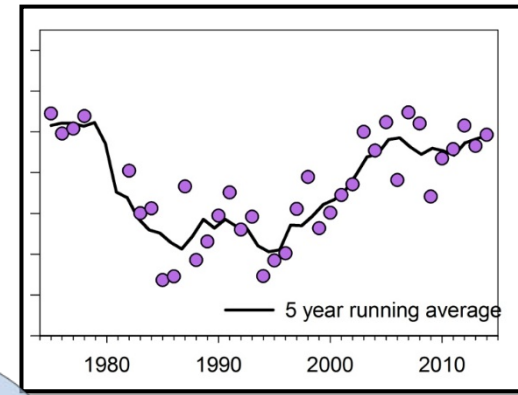
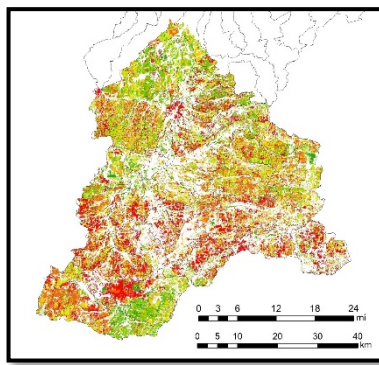


# Phosphorus

- Phosphorus is a macronutrient needed by all living things
- The average person contains 1.34 lbs of P
  - Bones, phospholipid bilayers, ATP, RNA, DNA, etc.
  - Humans tend to consume far more P than is needed (meaning we excrete a lot)
- The only primary source of P is from rocks and minerals- P fertilizer is mined!
- The most common form of P, phosphate, like to bind to iron and aluminum minerals
- Soil and sediment are great at removing P, but it can also accumulate and release for a long period of time

# Nitrogen

- Nitrogen is also a macronutrient needed by all living things
- The average person contains 5.4 lbs of N
  - Proteins and amino acids, ATP, RNA, DNA, etc.
  - Humans tend to consume far more N than is needed (meaning we excrete a lot)
- Nitrogen gas makes up 78% of Earth's atmosphere
- Microbes can breathe oxidized forms of N (i.e., nitrate) and return it to the atmosphere; therefore N rarely accumulates in freshwaters
- Nitrate-N is a drinking water contaminant that can impact the ability of hemoglobin to bind oxygen



*To generate knowledge about the dynamics of water and soil resources in order to improve water quality and availability*

# Heidelberg Tributary Loading Program

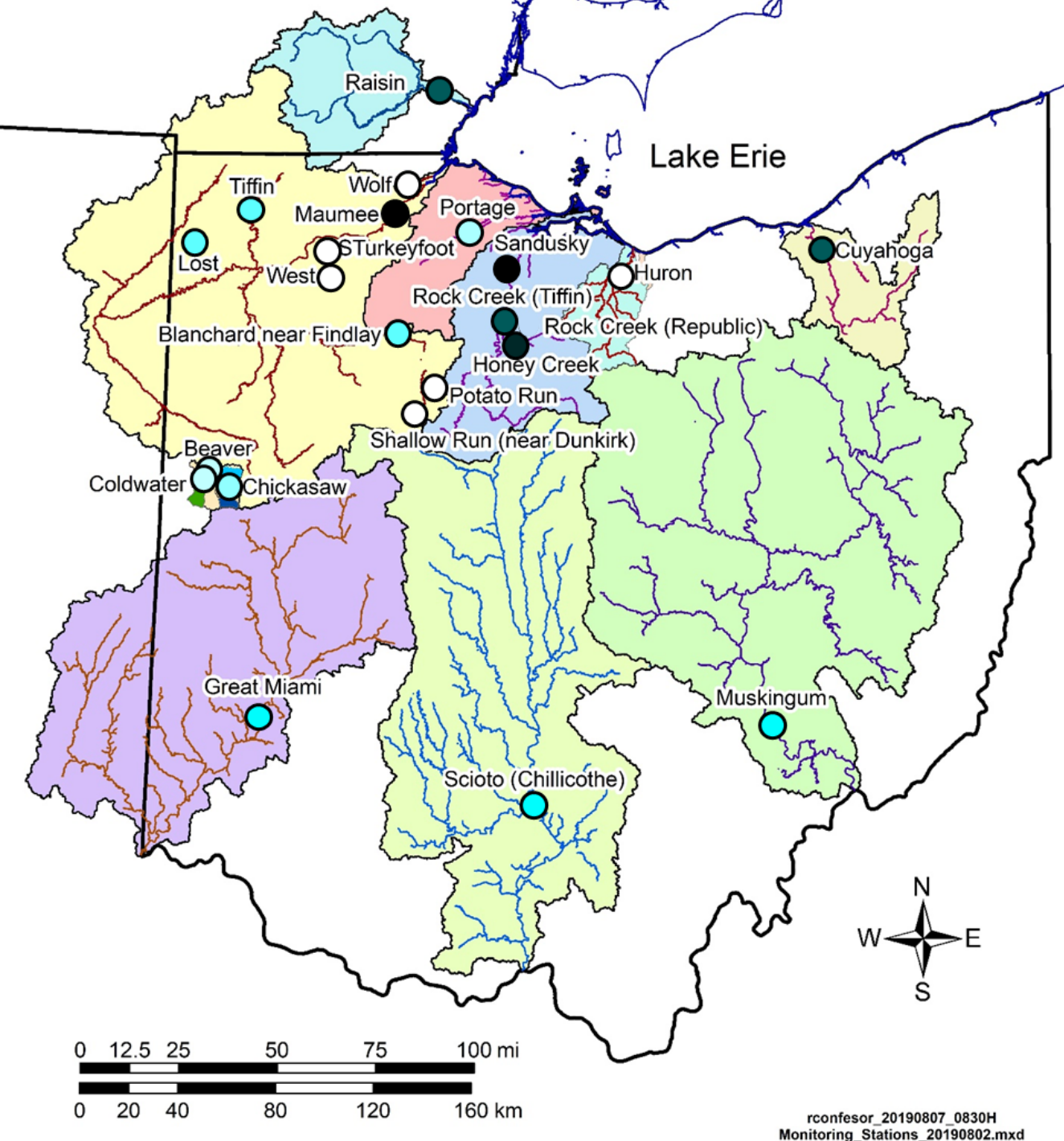
1975 1996 2018

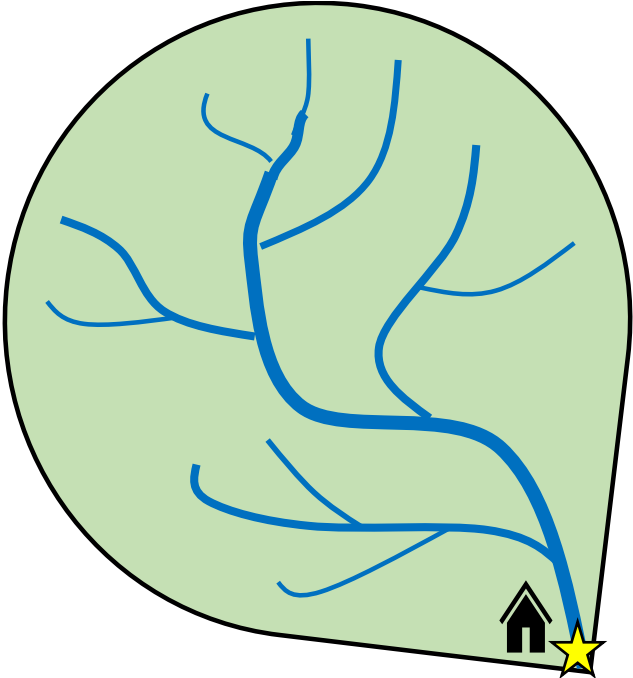
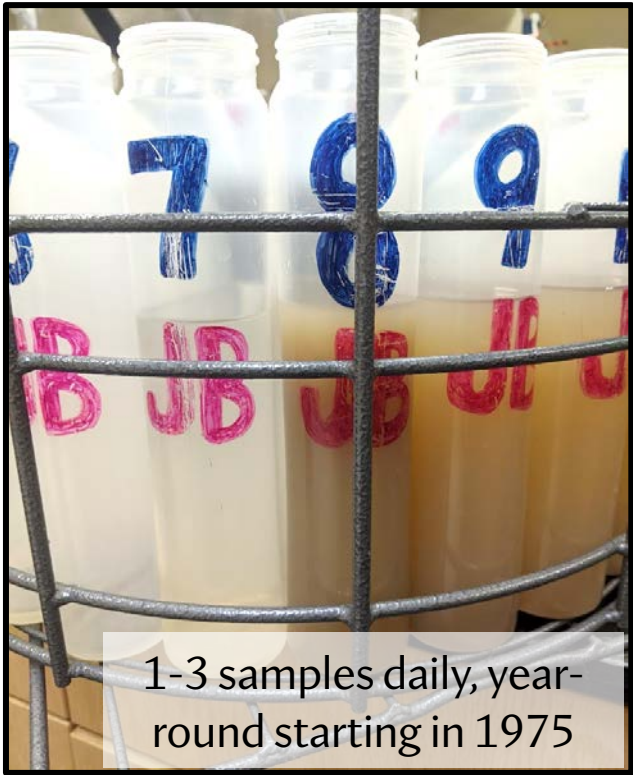
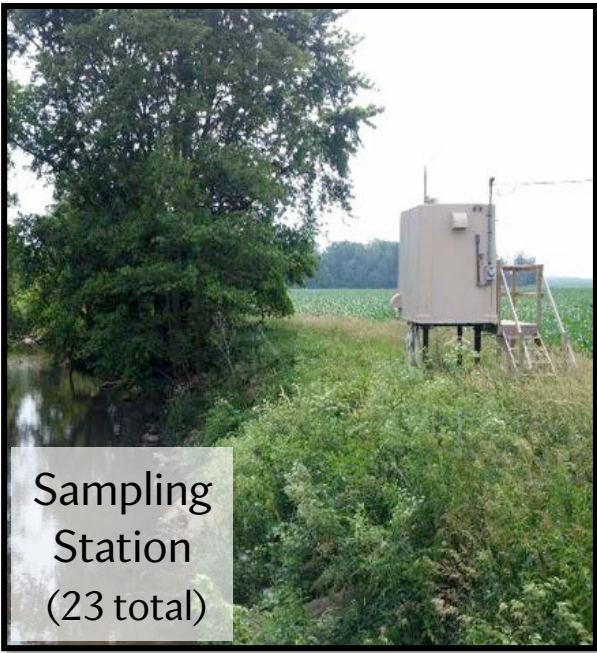
GOAL: quantify the loads (or mass) of contaminants from watersheds draining different land uses

We currently sample from 23 locations

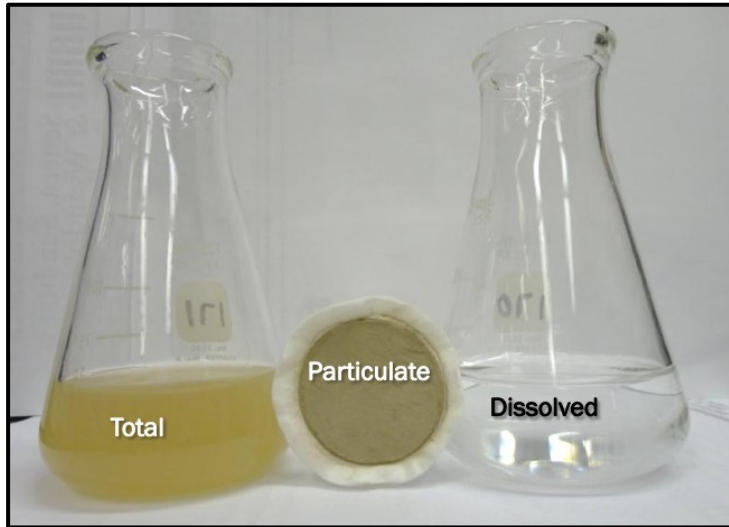
The HTLP began in 1974 with the Sandusky and Maumee

The newest stations are smaller watersheds to help test practice effectiveness

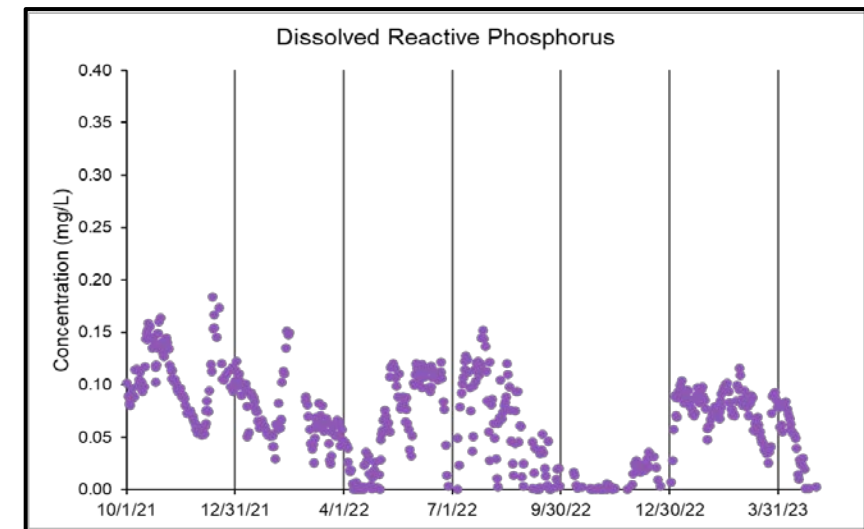




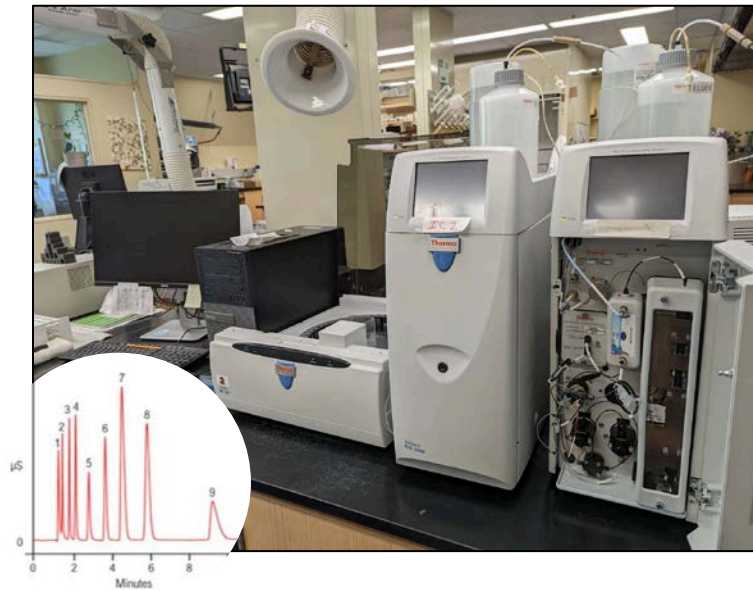
## Sample processing




## Data!



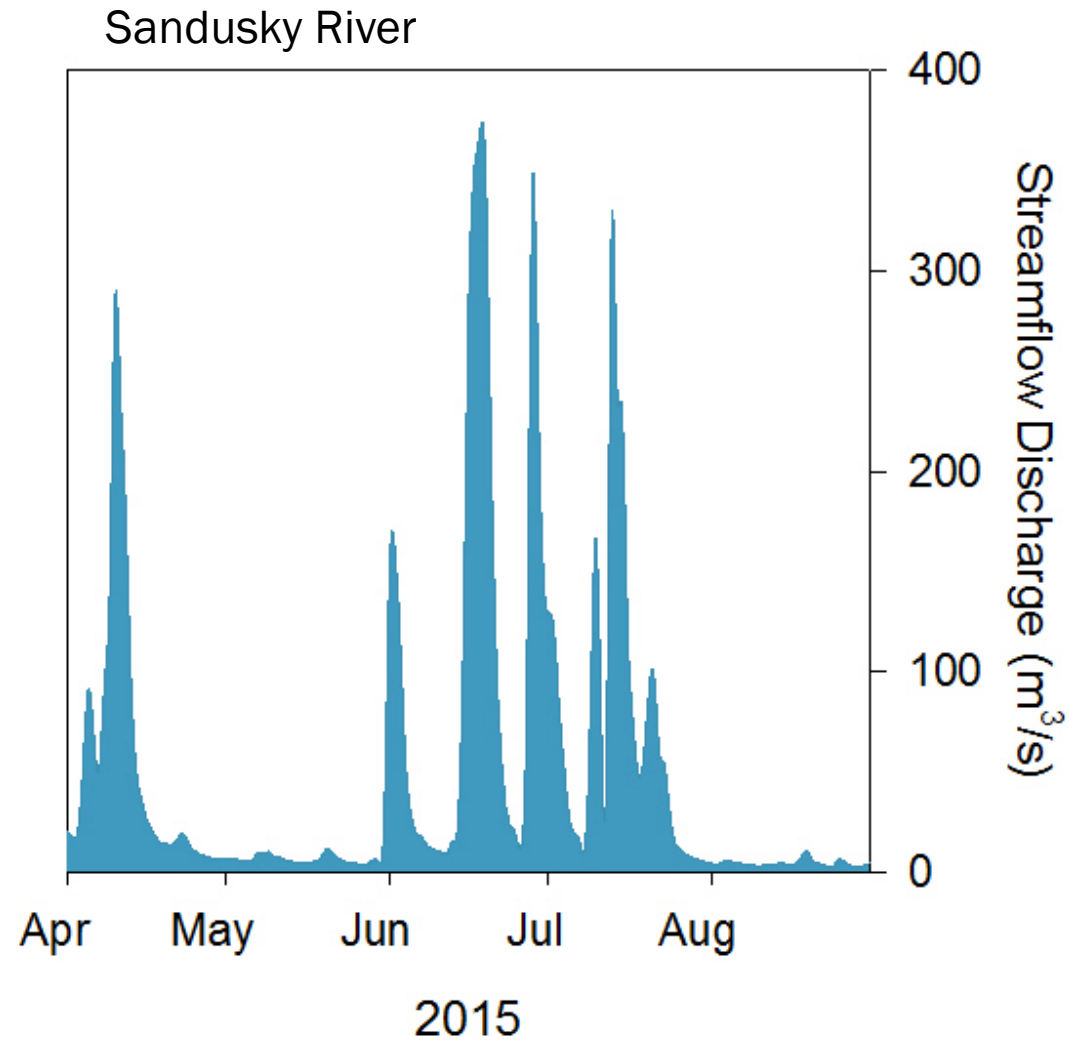
## Sample analysis





**Why do we need this approach to accurately  
assess sediments and nutrients in stormwater?**

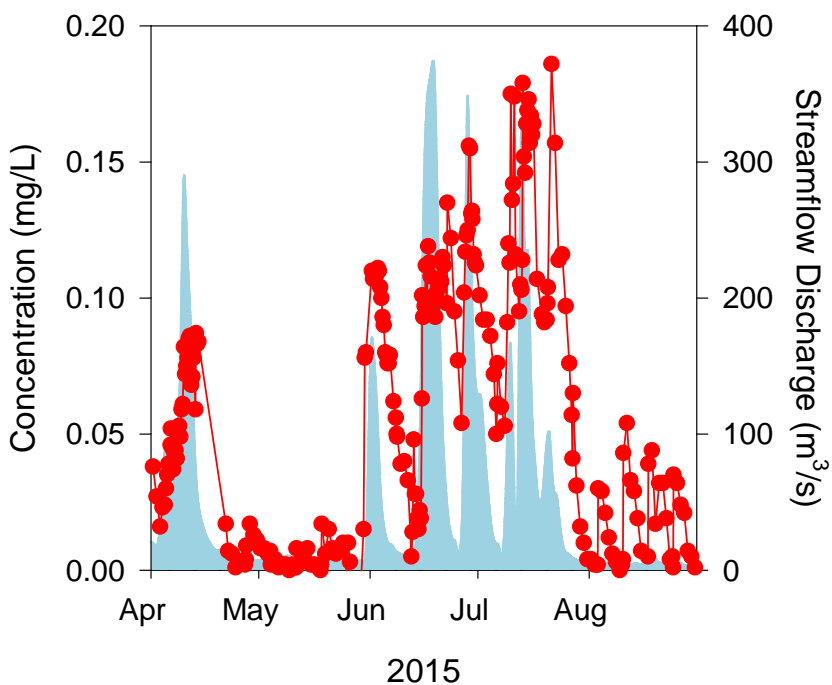
# Storms happen rapidly and can be difficult to predict



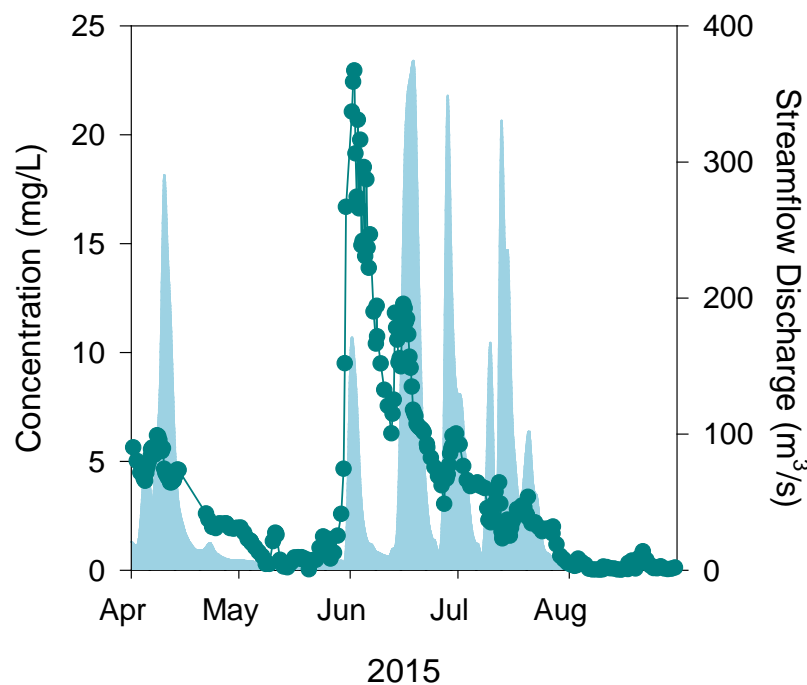
# Water pollution from land runoff increases with storms

## Sandusky River

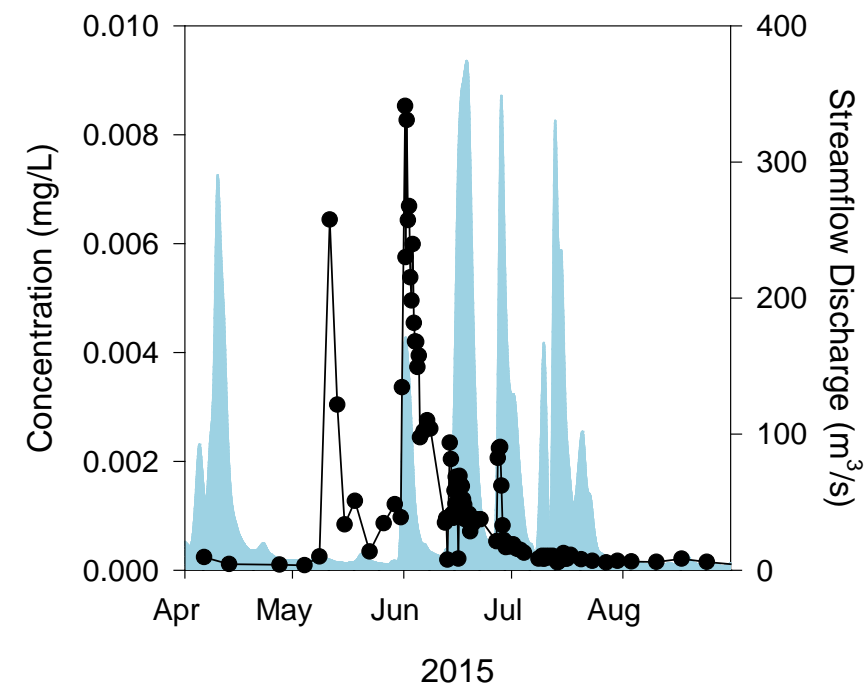
Dissolved Reactive Phosphorus



Nitrate-Nitrogen

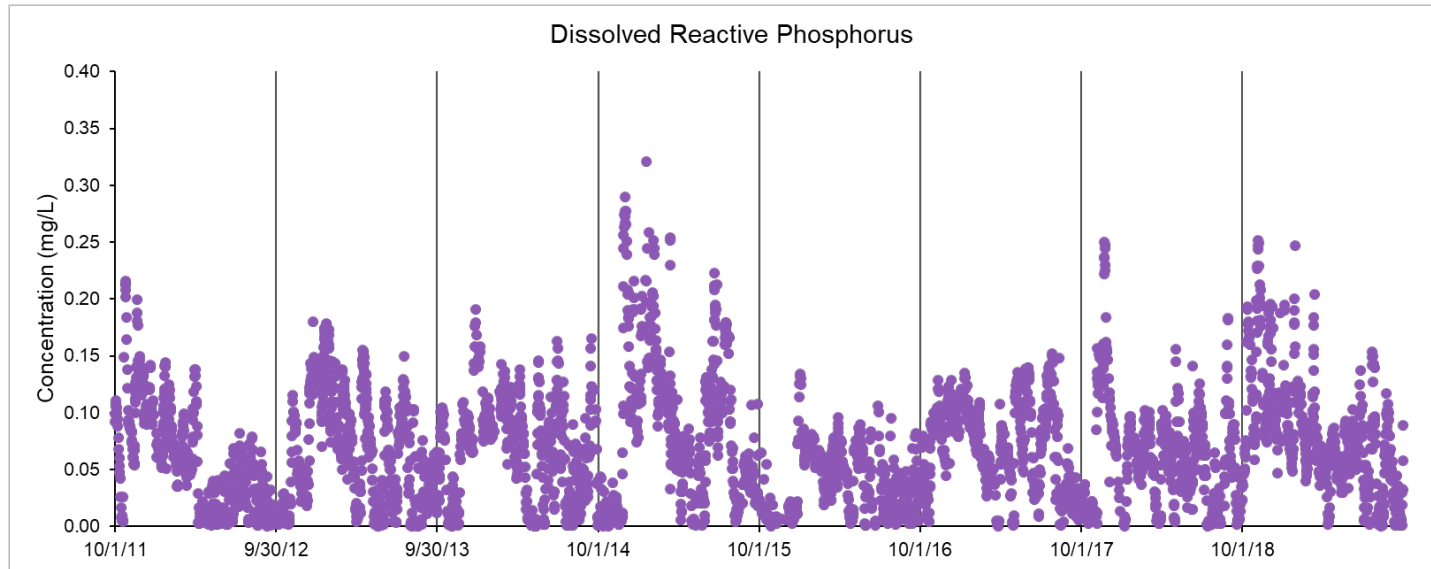


Atrazine

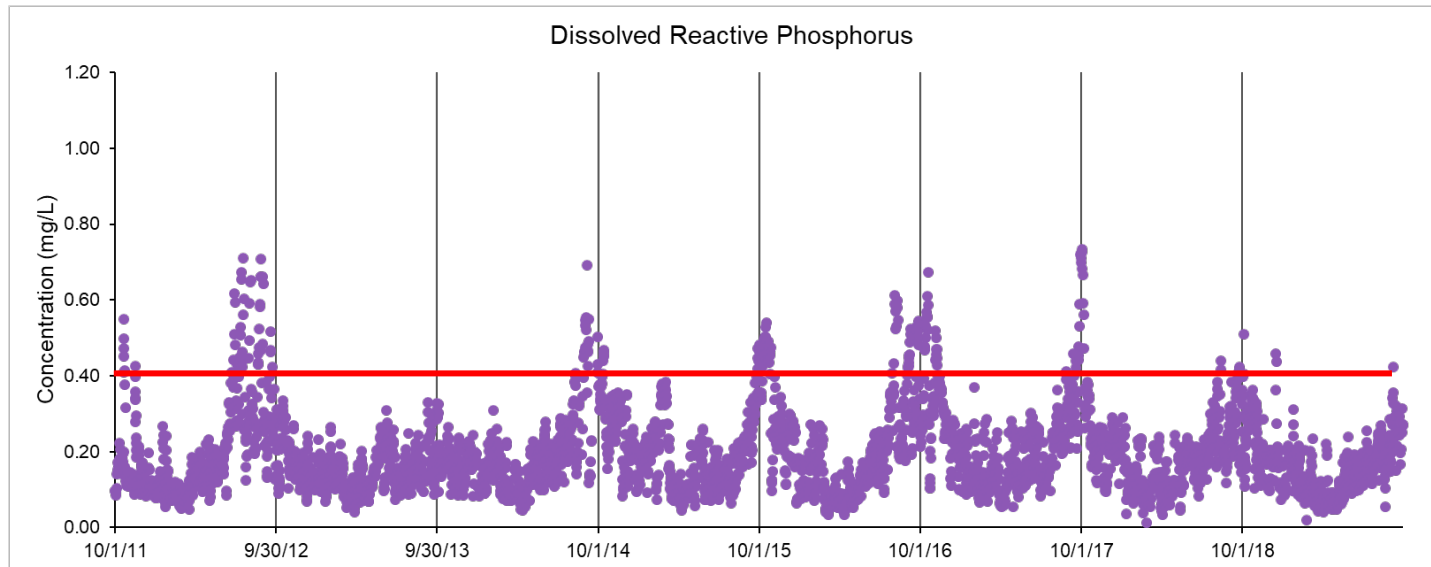


# The response of water pollution to storms varies by river

Maumee

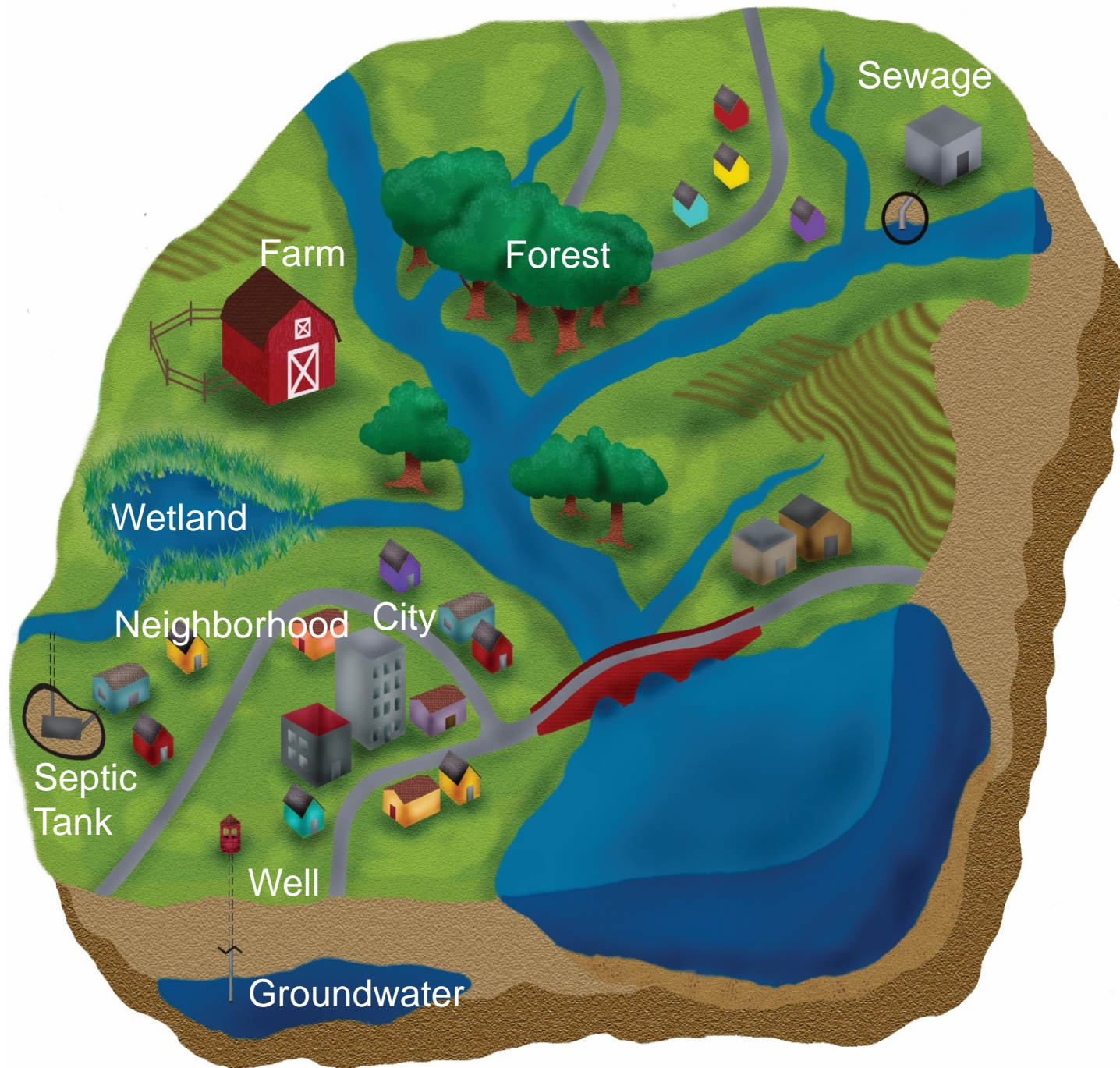


Blanchard





**How do nutrients enter our  
rivers and streams?**



# What is a watershed?

**an area of land that drains  
all the streams and rainfall  
to a common outlet**

# Point sources



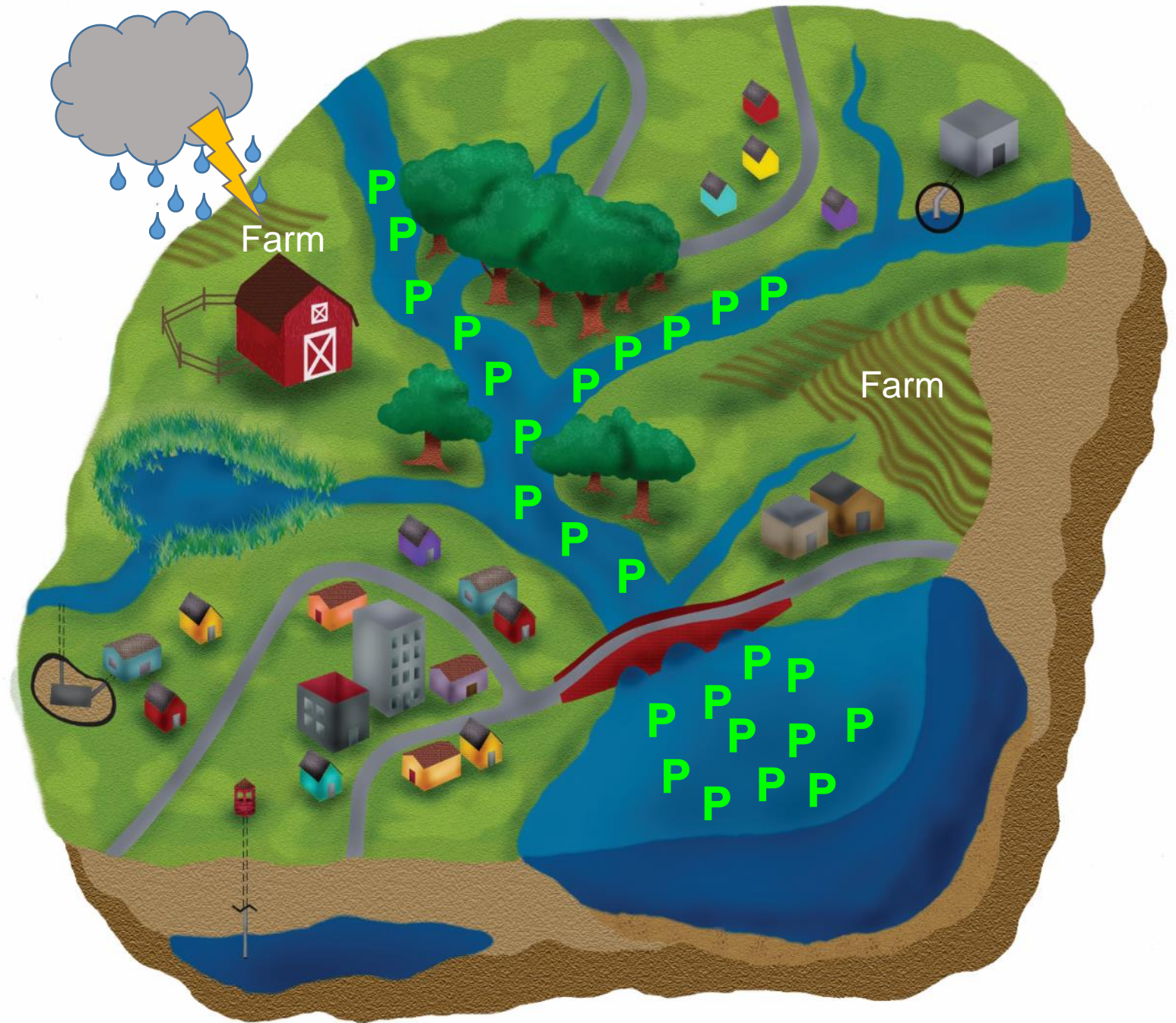
- Point sources tend to have higher concentrations during low flow
- Blooms from point sources tends to be localized near the input rather than at a distance



# Nonpoint sources









- Nonpoint sources tend to have higher concentrations during high flow
- Blooms from nonpoint sources tends to occur in the downstream receiving waterbodies where water velocity is slower (lakes, estuaries, stormwater ponds, reservoirs, etc)





# TIME TO BE A WATER Q DETECTIVE

| Location  | Sediment<br>(mg/L) | Phosphorus<br>( $\mu$ g/L) |
|---|--------------------|----------------------------|
|  up                      |                    |                            |
|  sewage                  |                    |                            |
|  farm/forest             |                    |                            |
|  septic/<br>neighborhood |                    |                            |
|  wetland                |                    |                            |
|  all                   |                    |                            |









# WHAT IS THE SOURCE OF THE HIGHEST CONCENTRATION?

| Location             | Sediment (mg/L) | Phosphorus (µg/L) |
|----------------------|-----------------|-------------------|
| up                   | 10              | 10                |
| sewage               | 15              | 800               |
| farm/forest          | 40              | 50                |
| septic/ neighborhood | 40              | 100               |
| wetland              | 5               | 5                 |
| all                  | 25              | 200               |









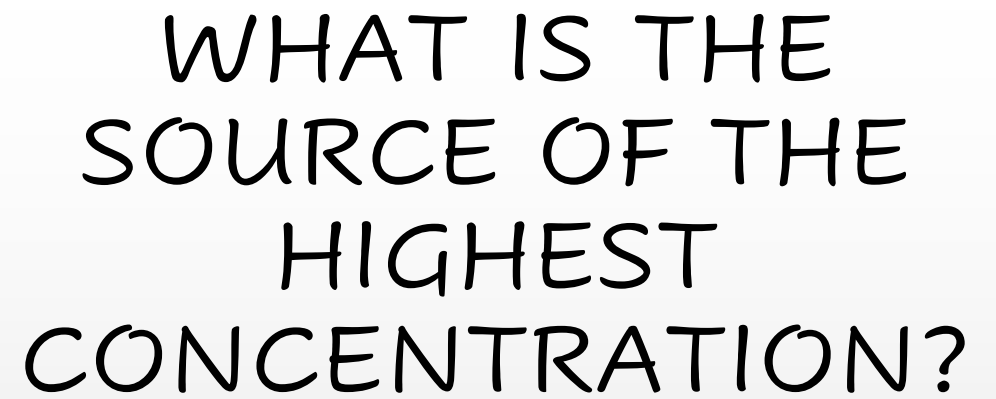
# WHAT IS THE SOURCE OF THE HIGHEST CONCENTRATION?







|   | Location             | Sediment (mg/L) | Phosphorus (µg/L) |
|---|----------------------|-----------------|-------------------|
|    | up                   | 10              | 10                |
|    | sewage               | 15              | 800               |
|    | farm/forest          | 40              | 50                |
|    | septic/ neighborhood | 40              | 100               |
|   | wetland              | 5               | 5                 |
|  | all                  | 25              | 200               |



# TIME TO BE A WATER Q DETECTIVE

| Location  | Sediment<br>(mg/L) | Phosphorus<br>( $\mu\text{g/L}$ ) |
|---|--------------------|-----------------------------------|
|  up                      |                    |                                   |
|  sewage                  |                    |                                   |
|  farm/forest             |                    |                                   |
|  septic/<br>neighborhood |                    |                                   |
|  wetland                |                    |                                   |
|  all                   |                    |                                   |

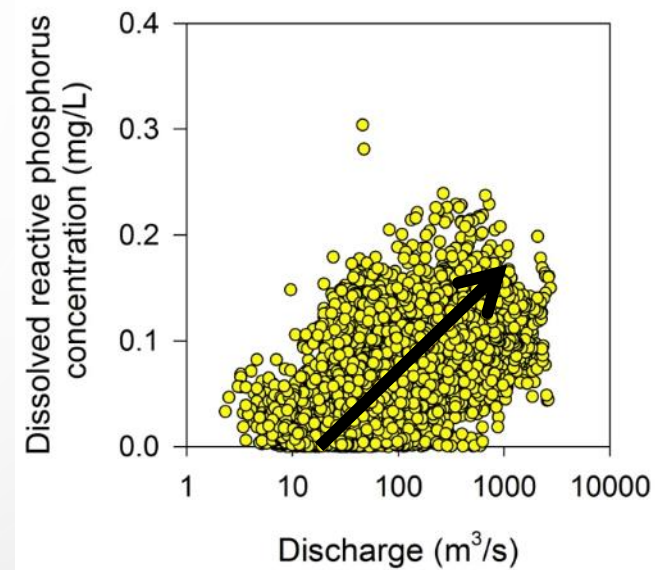
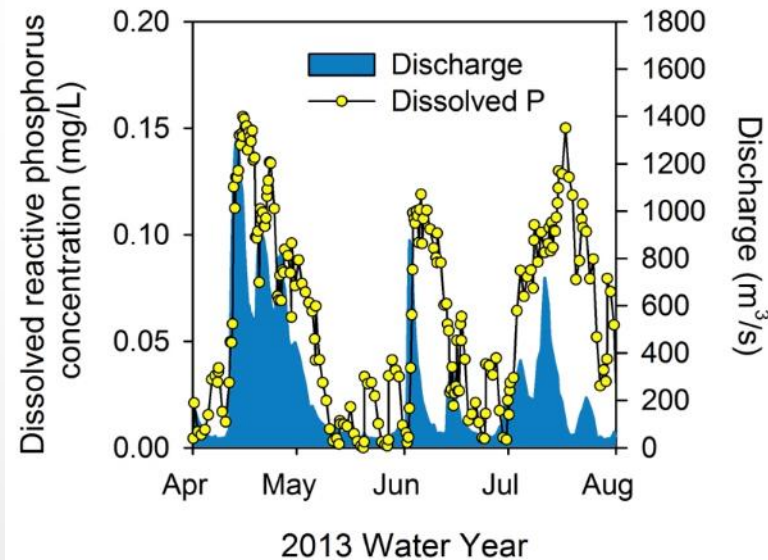


|   | Location                | Sediment<br>(mg/L) | Phosphorus<br>( $\mu\text{g/L}$ ) |
|---|-------------------------|--------------------|-----------------------------------|
|    | up                      | 100                | 50                                |
|    | sewage                  | 100                | 55                                |
|    | farm/forest             | 750                | 200                               |
|    | septic/<br>neighborhood | 300                | 55                                |
|   | wetland                 | 30                 | 15                                |
|  | all                     | 450                | 150                               |

# HOW DOES PHOSPHORUS ENTER SURFACE WATER?

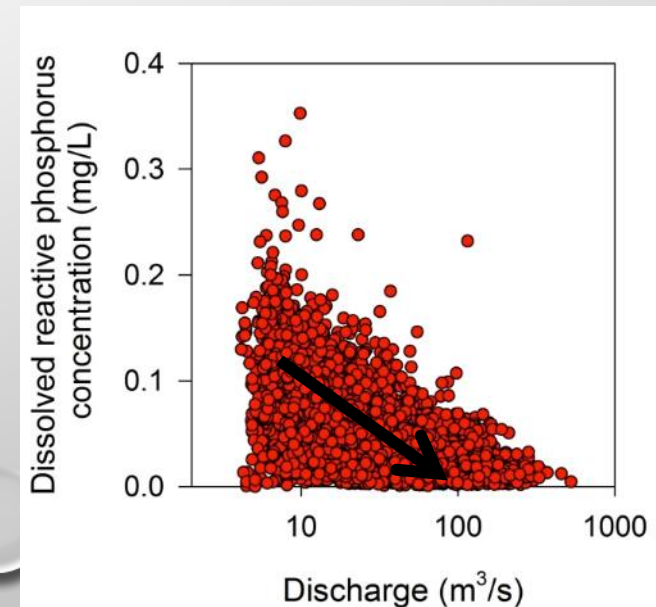
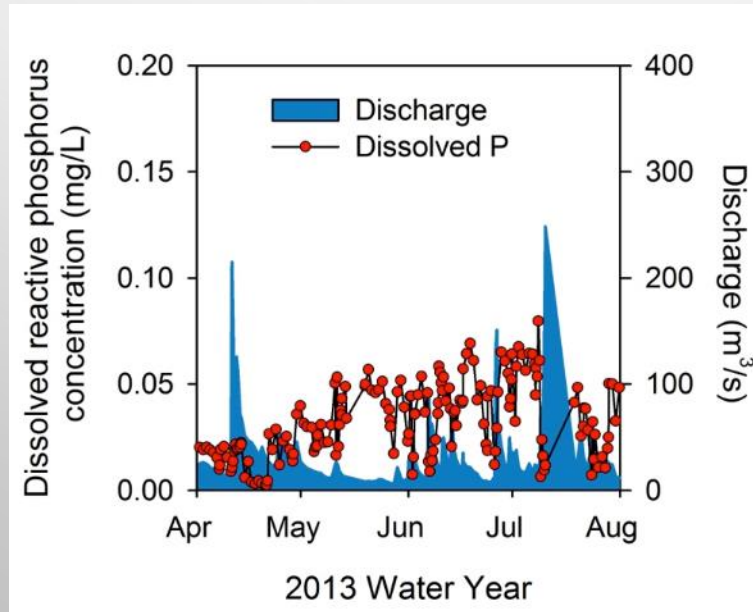
## Maumee River: Nonpoint Sources

- P inputs dominated by agricultural runoff
- Concentrations increase with flow



## Cuyahoga River: Point Sources

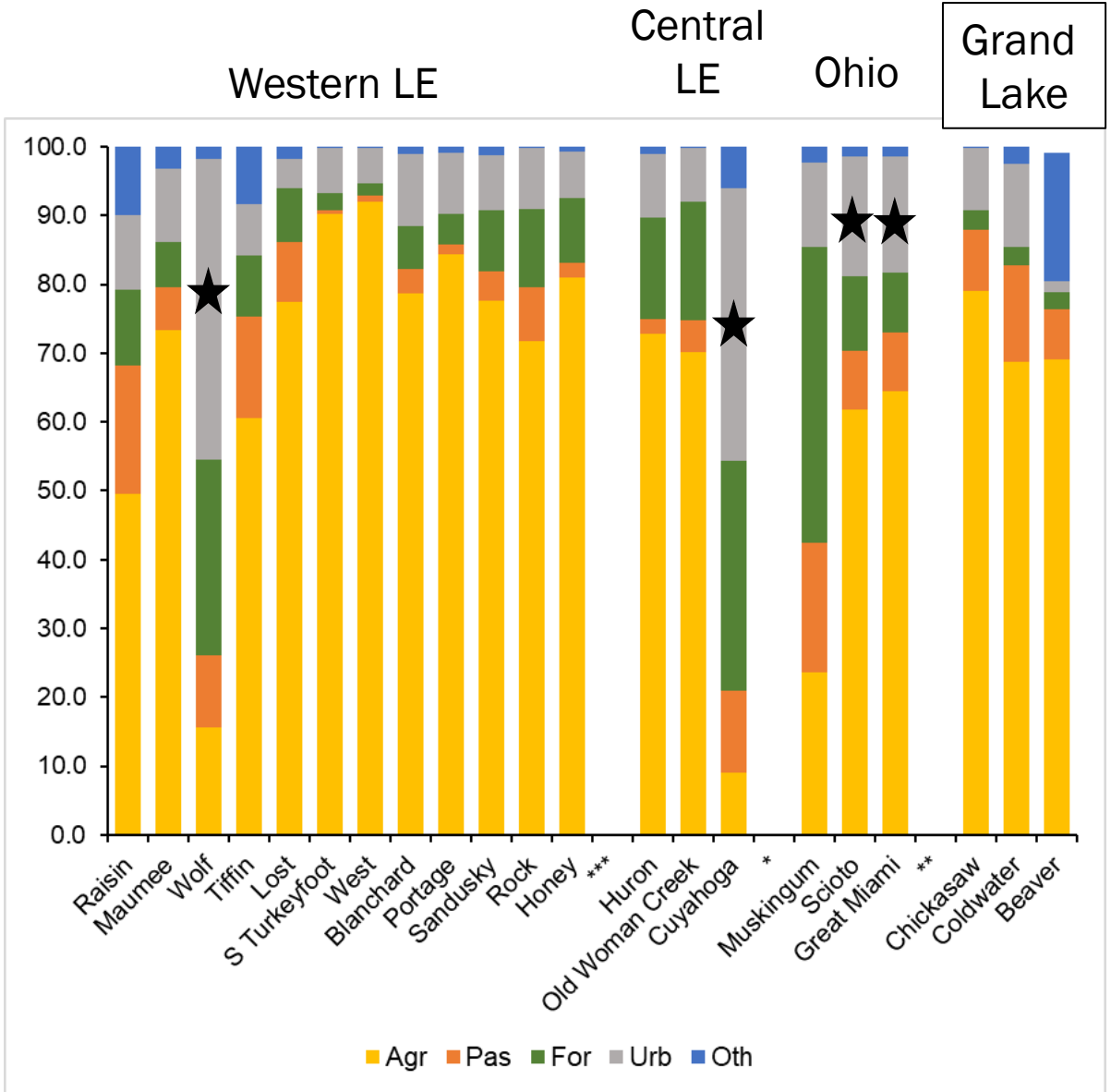
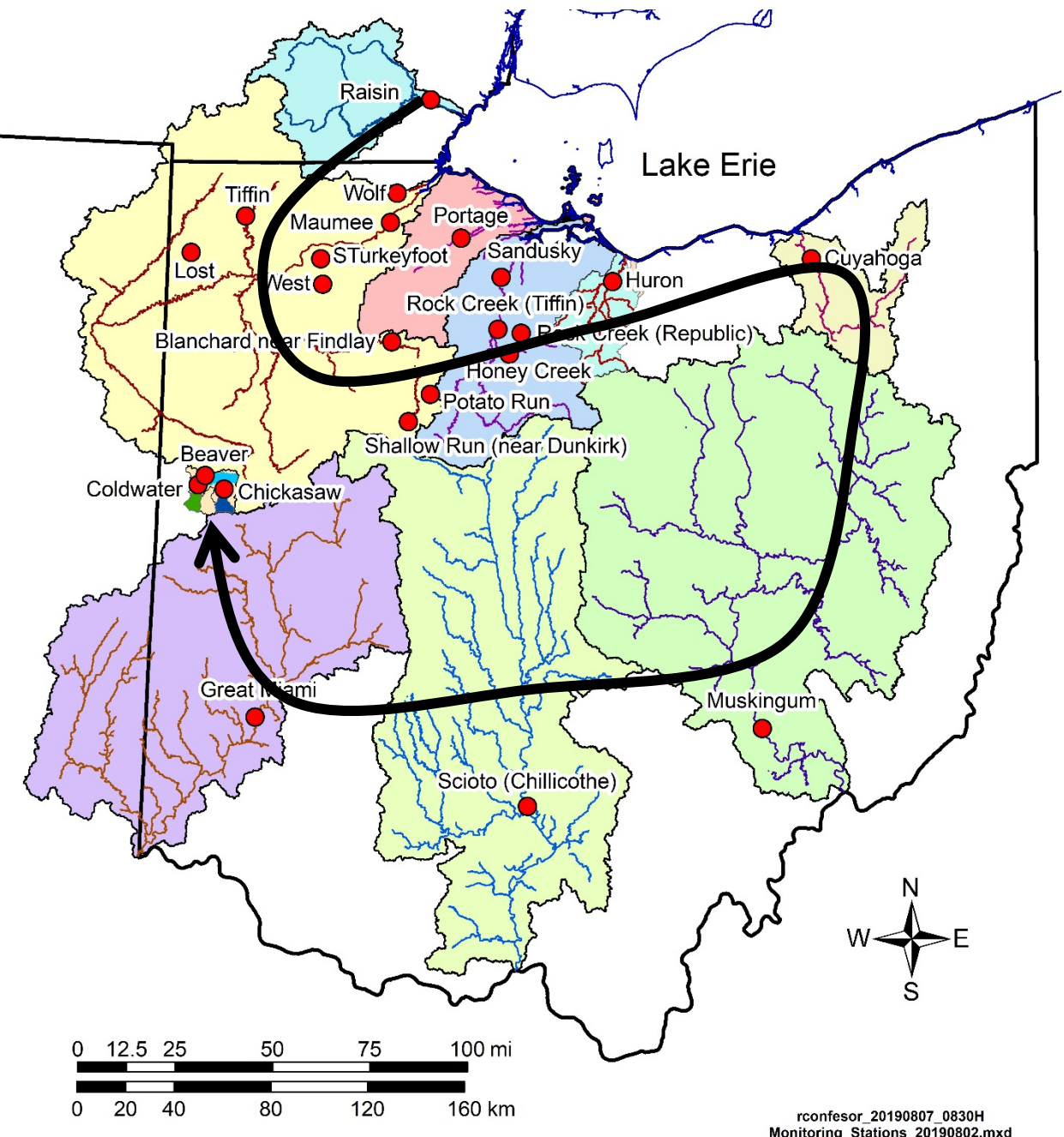
- P inputs dominated by sewage effluent
- Concentrations decrease with flow





**How do phosphorus loads vary  
by watershed?**

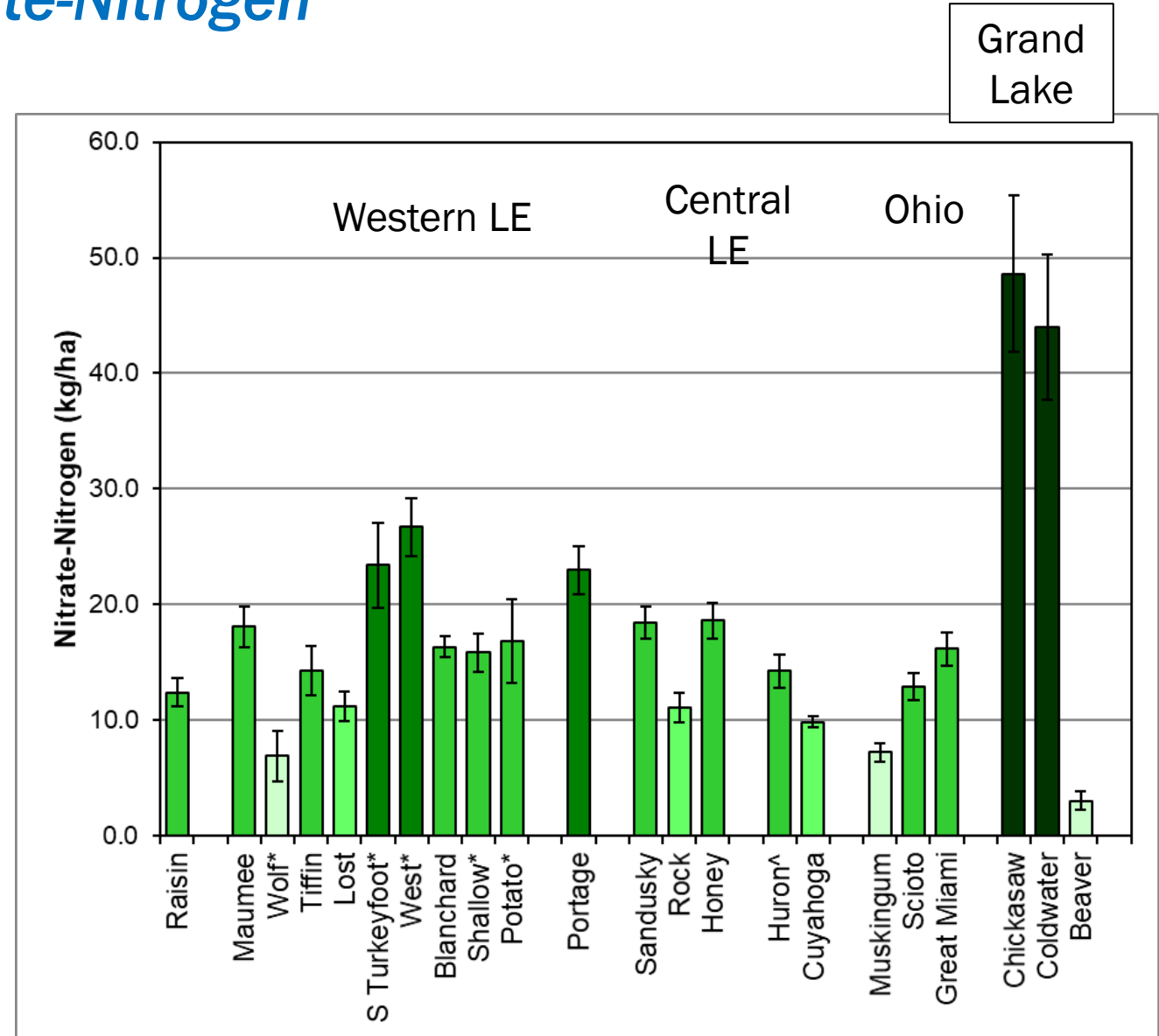
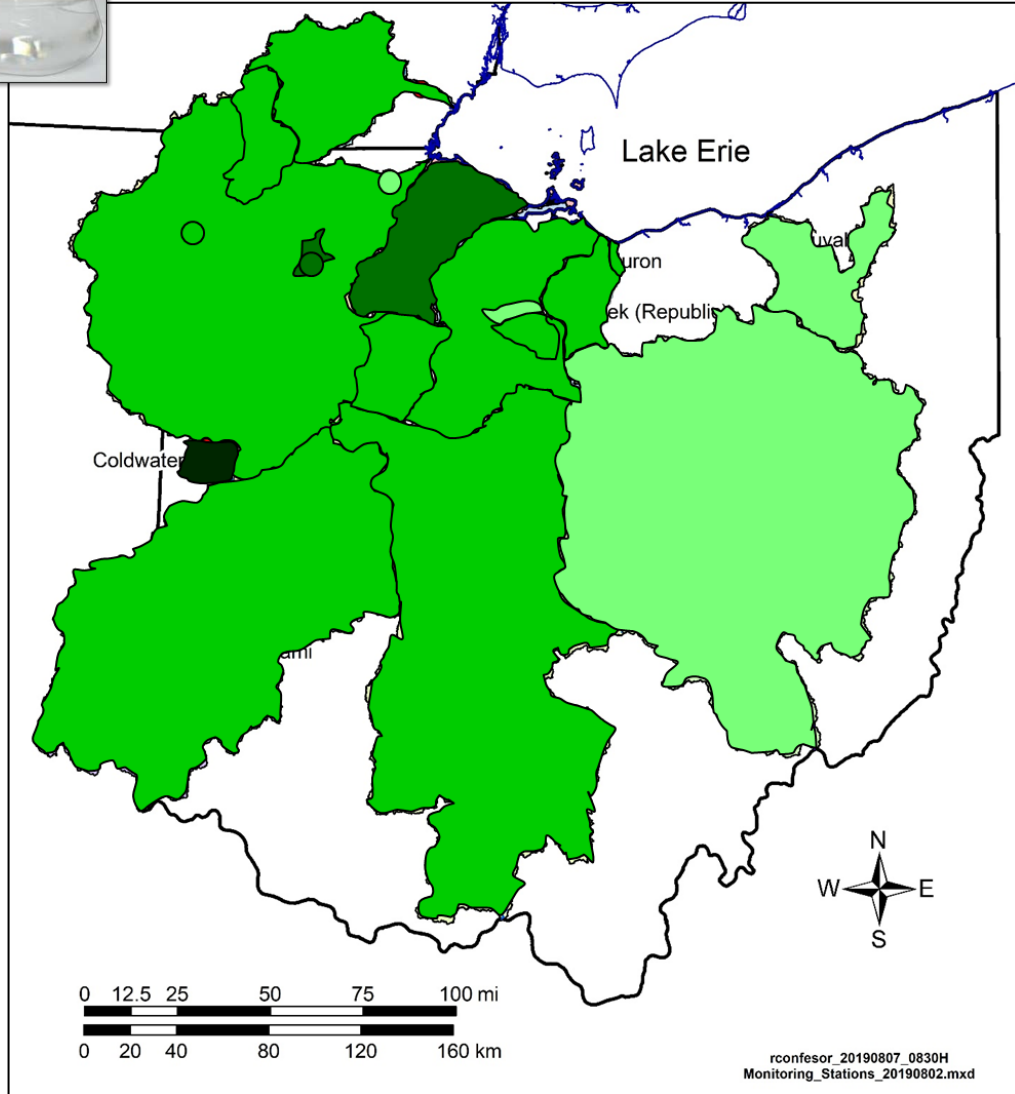
# Watershed Land Use





# Comparing watersheds: Unit Area Loads

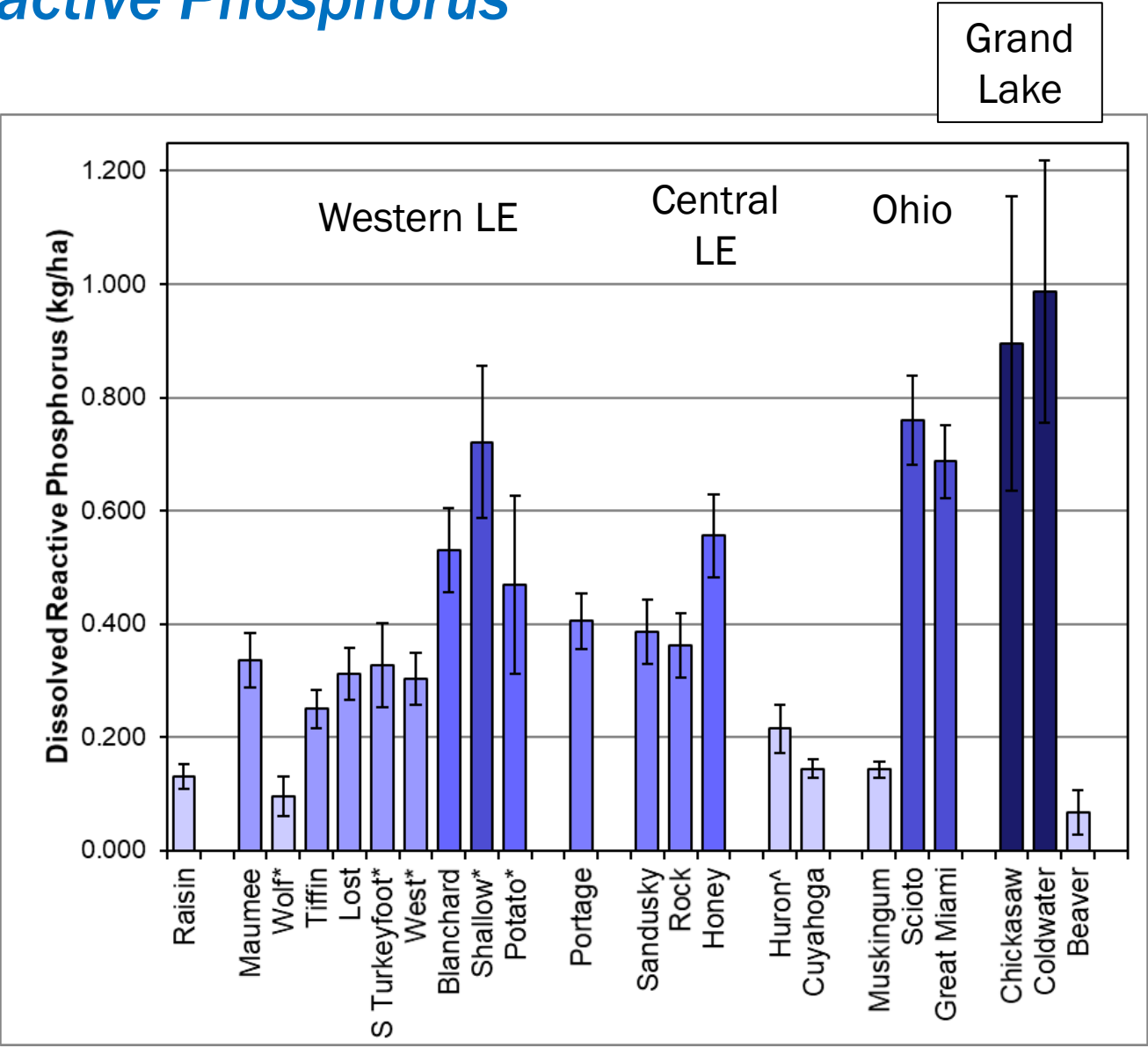
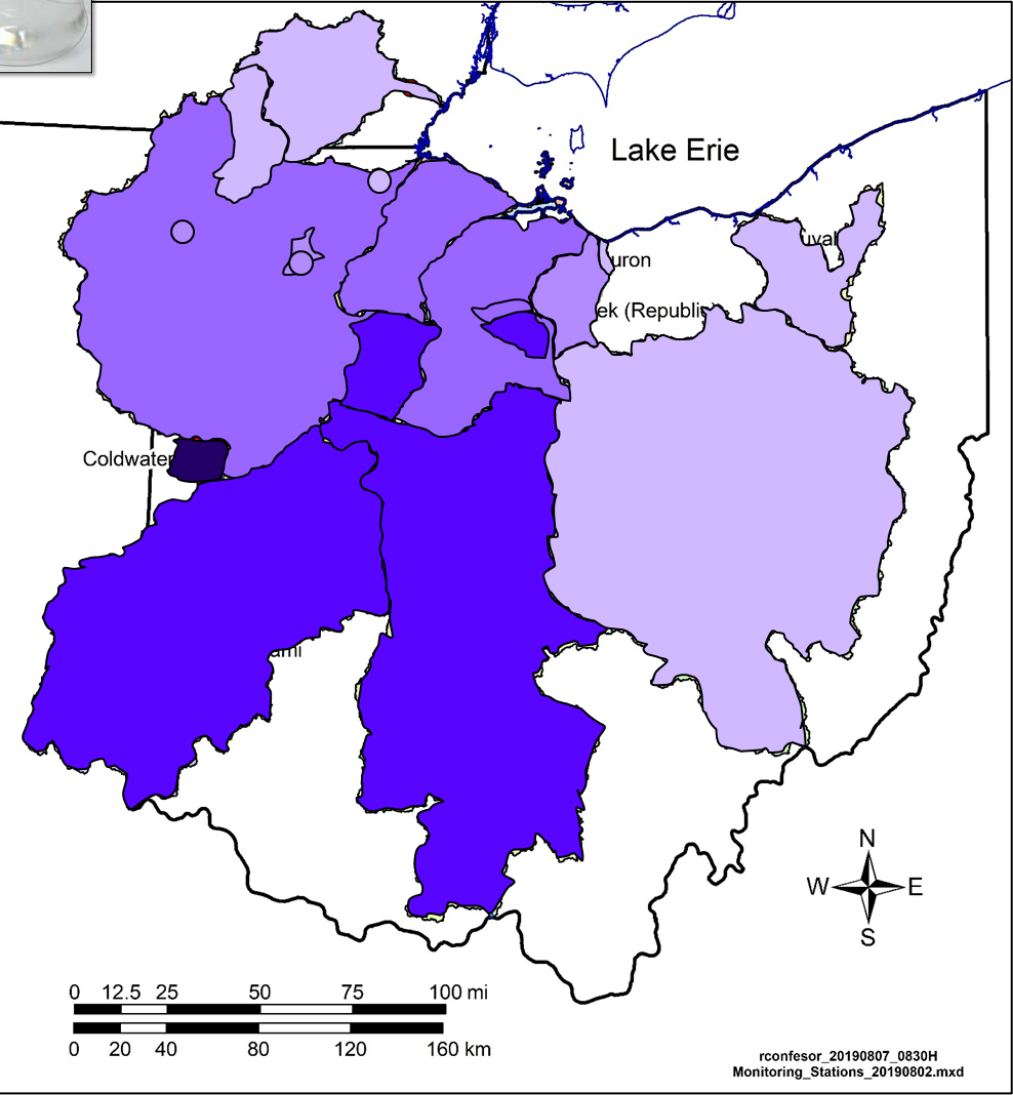
## Nitrate-Nitrogen



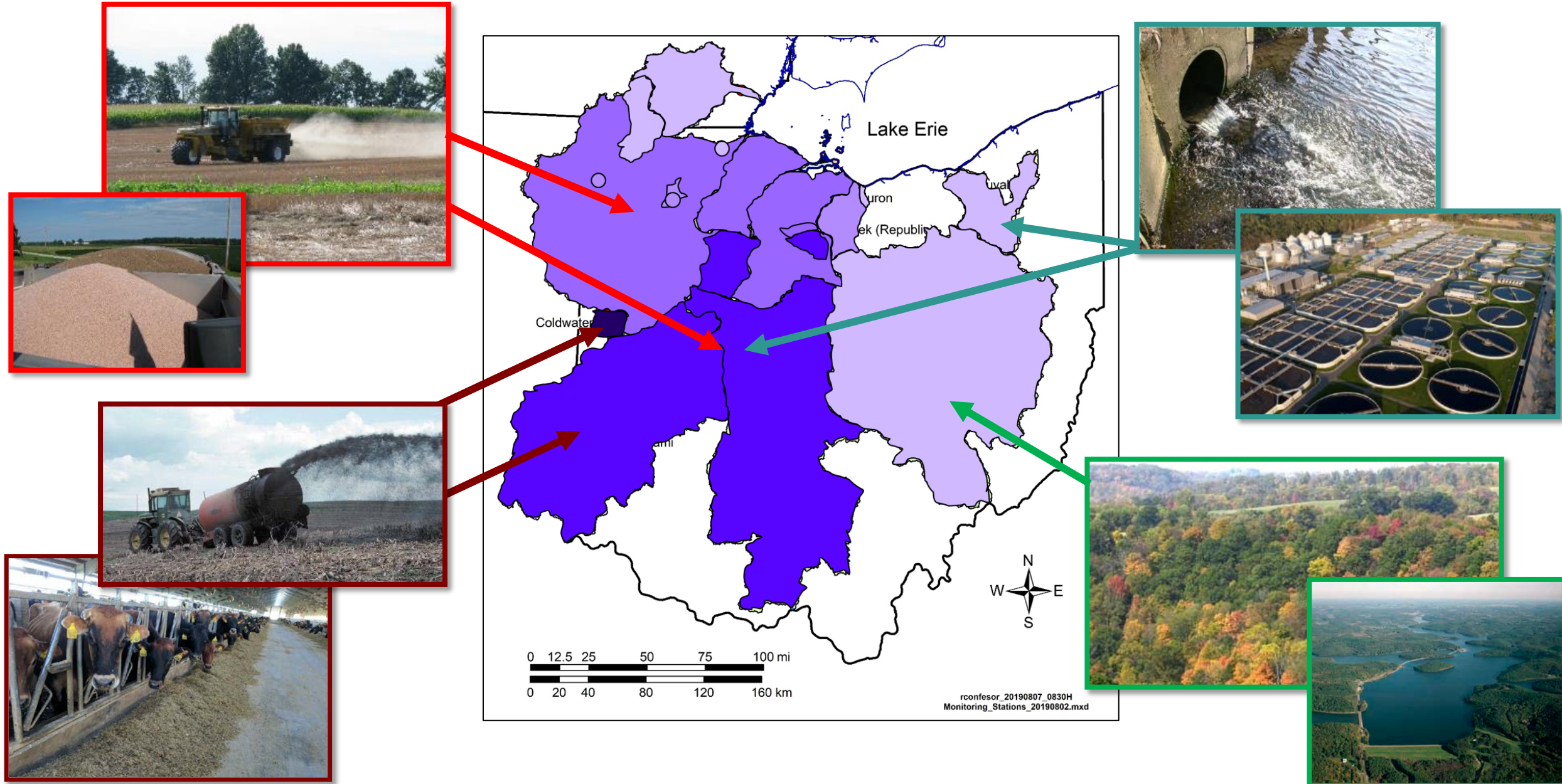


# Comparing watersheds: Unit Area Loads

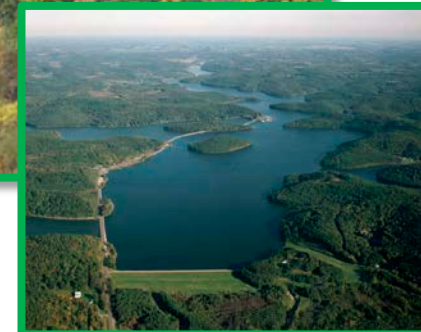
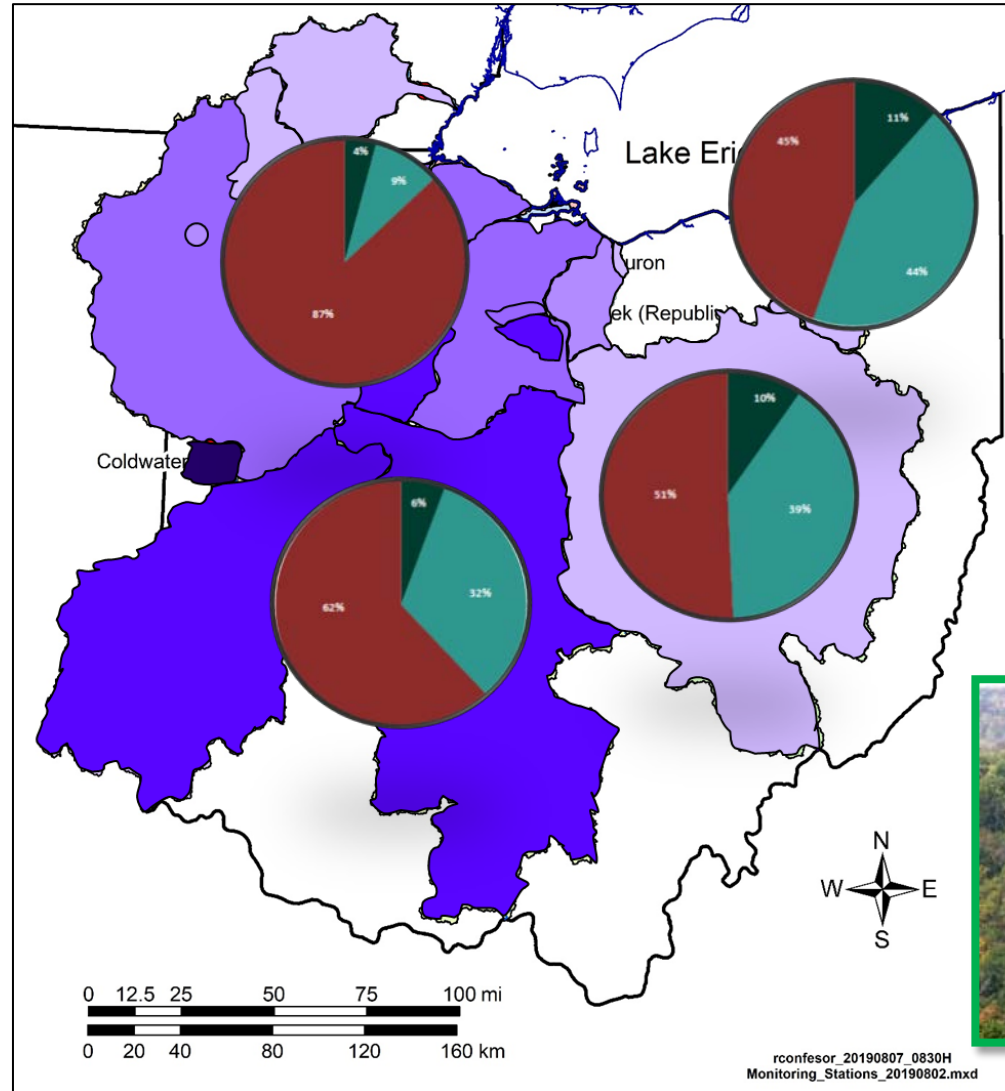
## *Dissolved Reactive Phosphorus*



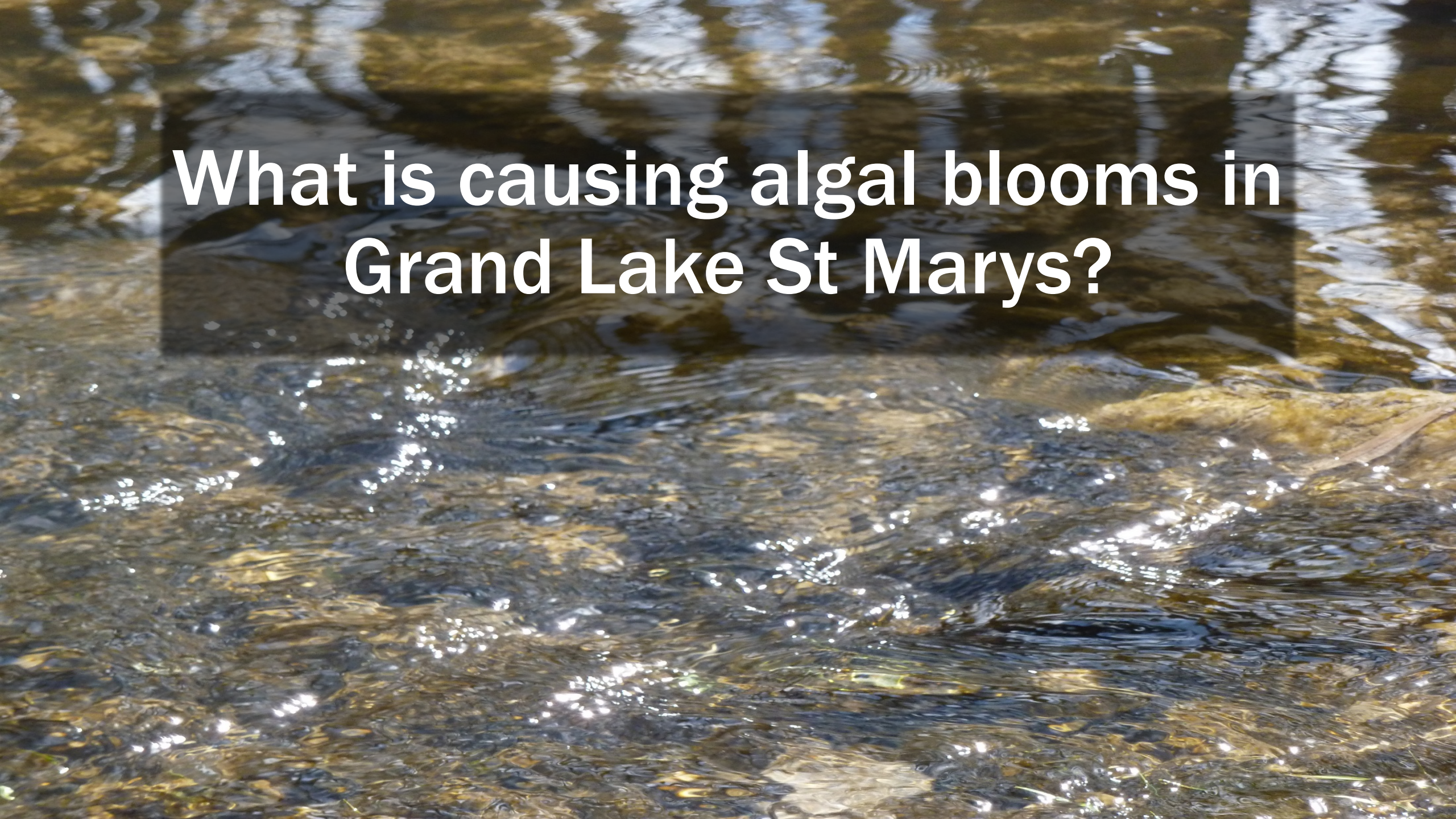
# Why are there differences across the watersheds?



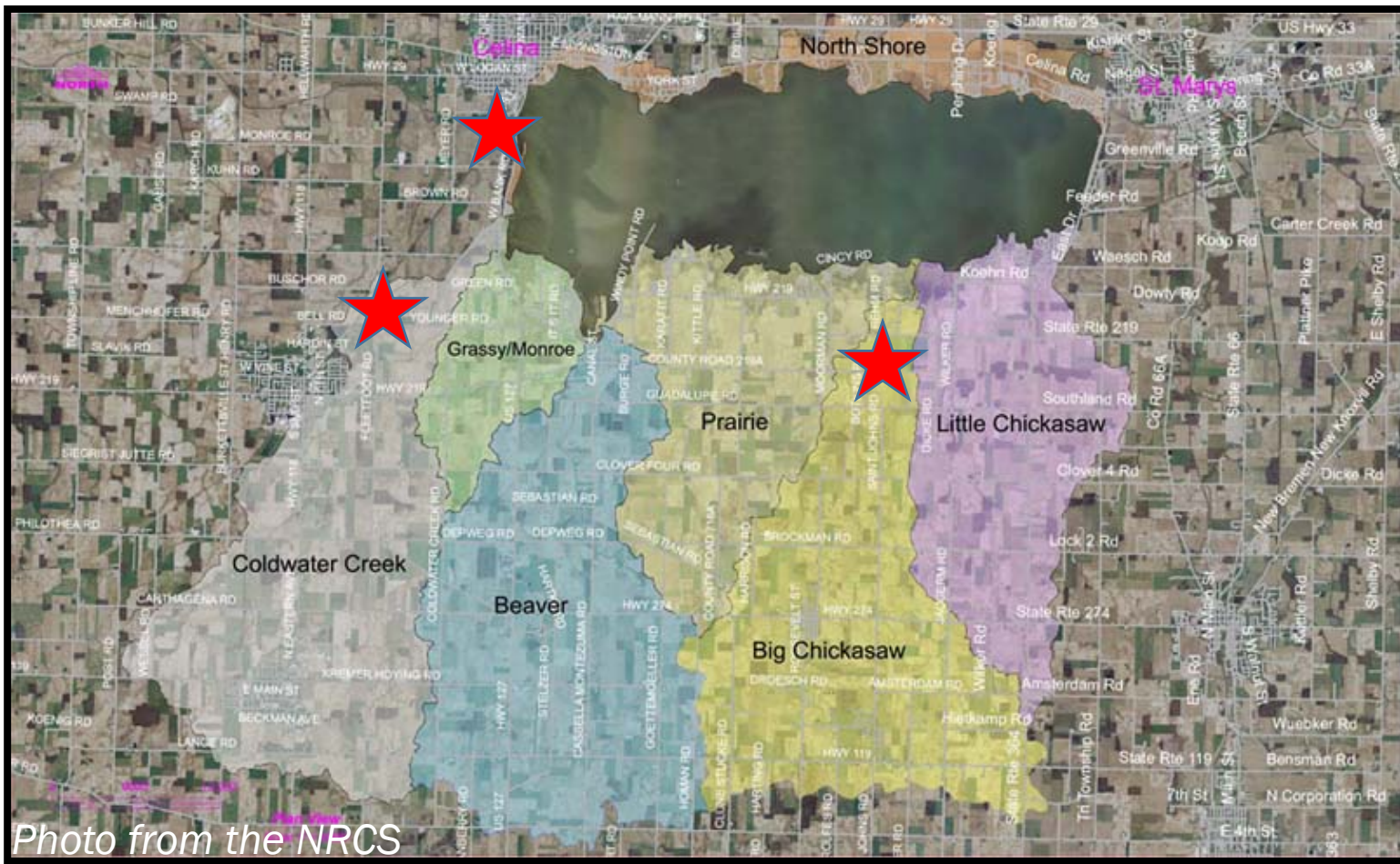
# Why are there differences across the watersheds?



SEPTIC POINT NONPOINT



**What is causing algal blooms in  
Grand Lake St Marys?**

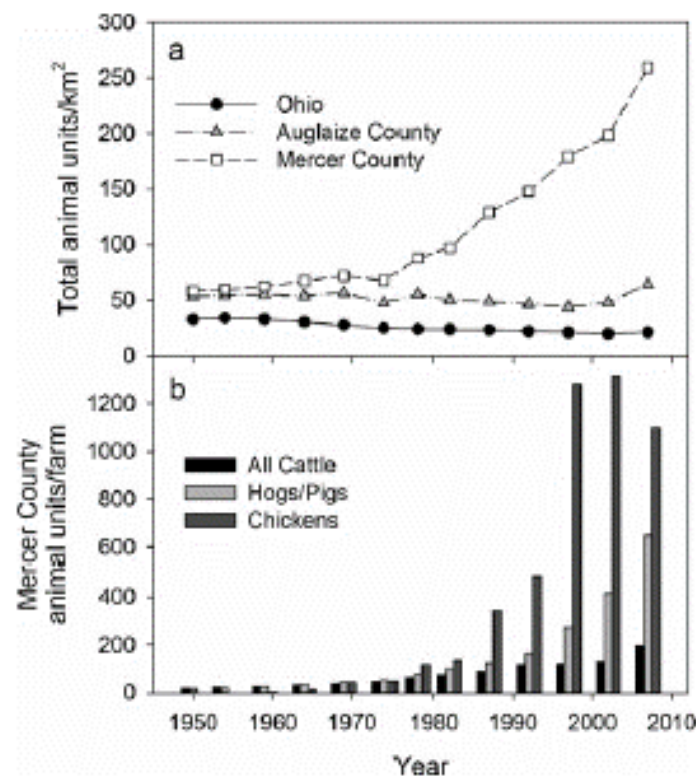


Chickasaw started in 2008  
 Coldwater started in 2012  
 Beaver (outflow) started in 2013

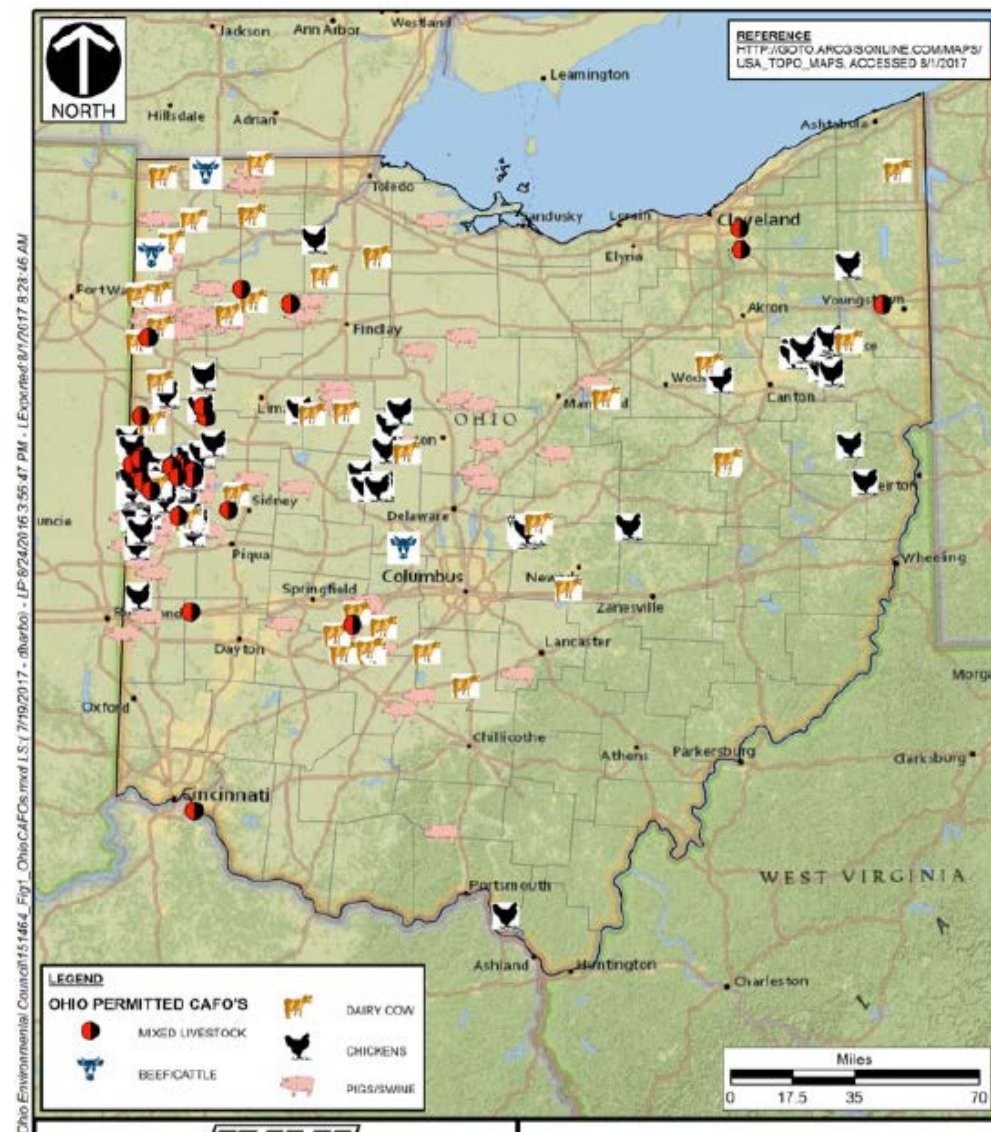


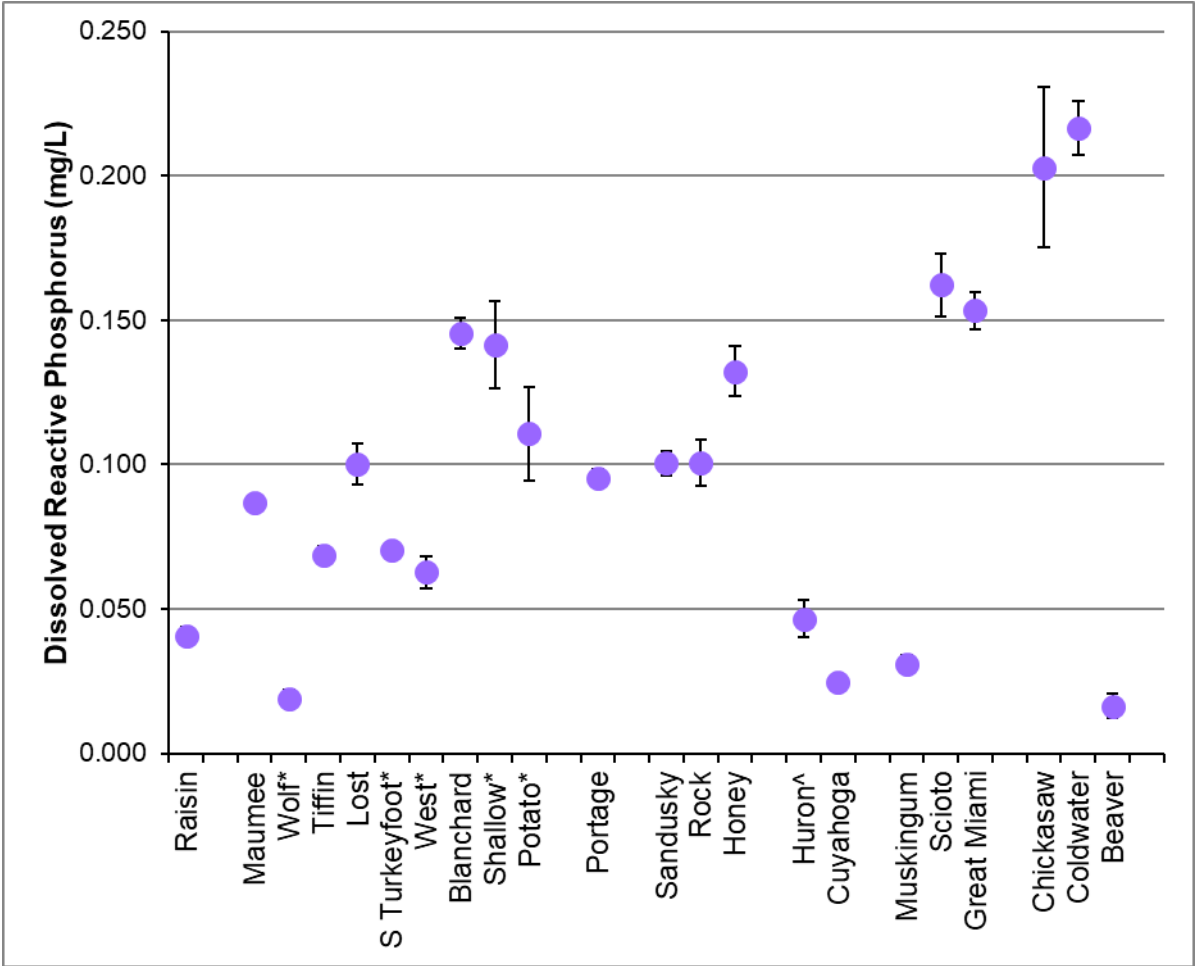
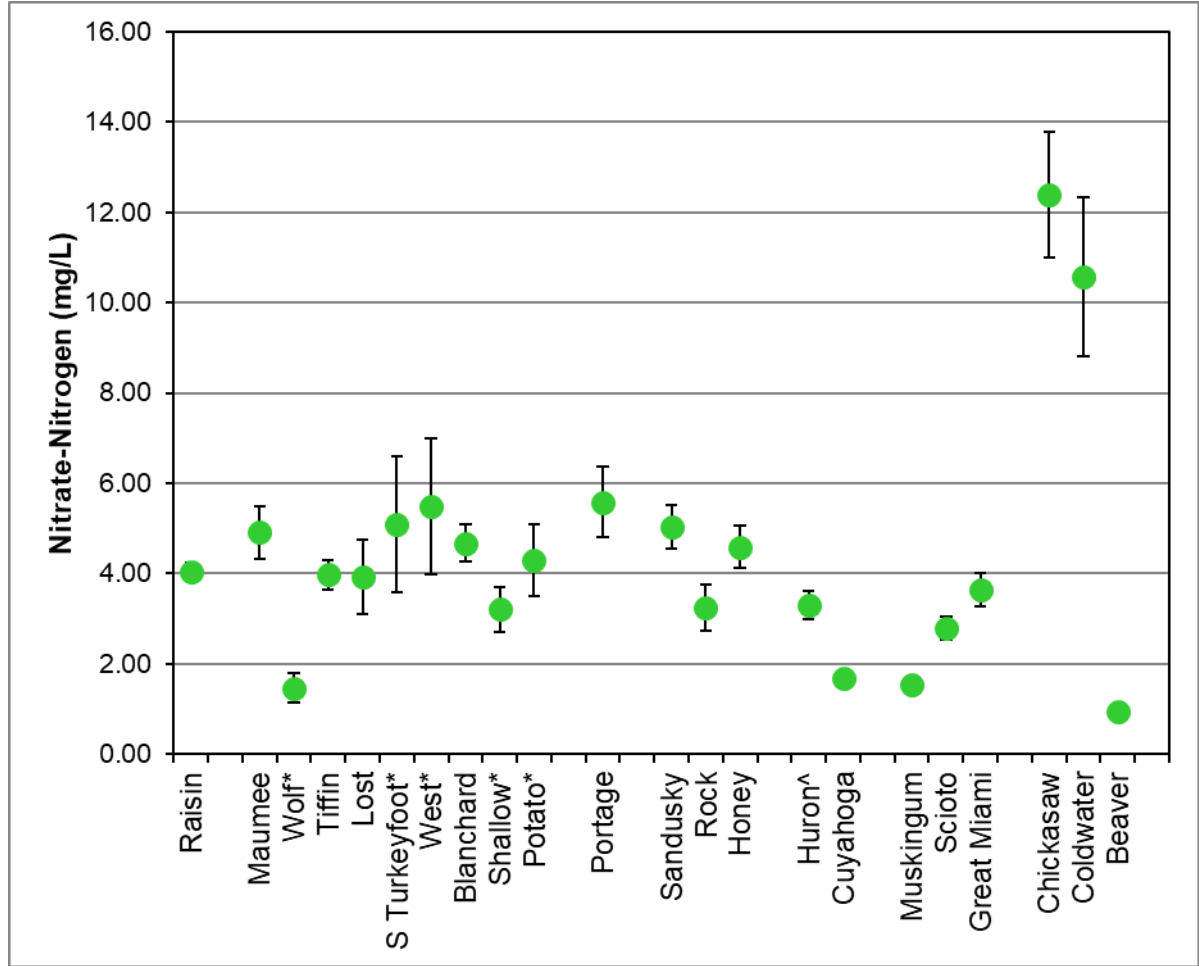
# Grand Lake St Marys

- 2007 National Lakes Assessment
  - Of 1252 lakes sampled, GLSM was in 97<sup>th</sup> percentile for chlorophyll a, and 3<sup>rd</sup> highest microcystin levels
- Why all the problems?
  - Lots of livestock in the basin
- Classified as a distressed watershed in 2011
- 99% of livestock producers and 90% of the cropland has a nutrient management plan
- ~20% of the watershed has received EQIP funding for cover crops
- No manure application after Dec 15<sup>th</sup> (in effect 2013) without approval to Mar 1<sup>st</sup>



From Filbrun et al. 2013

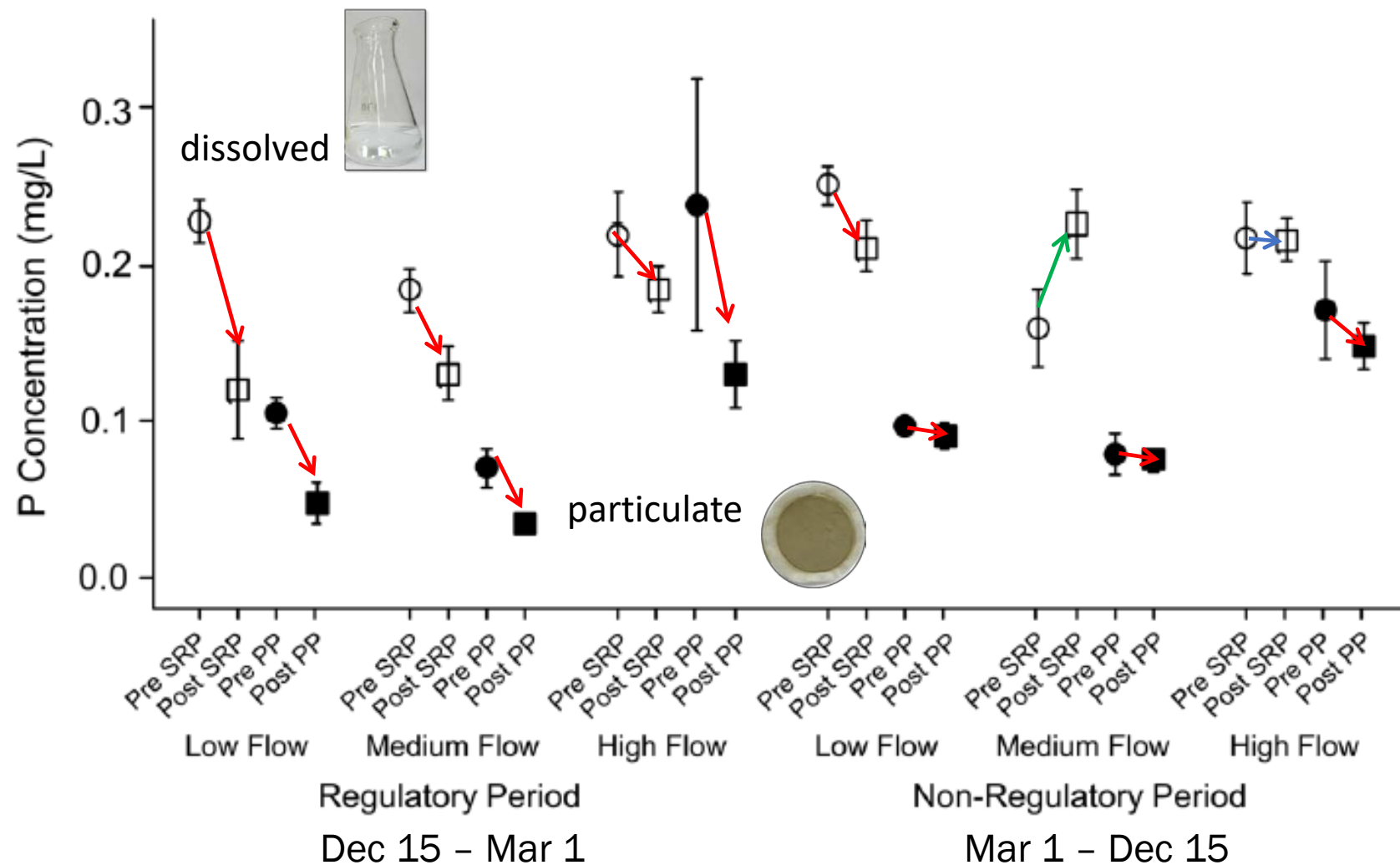




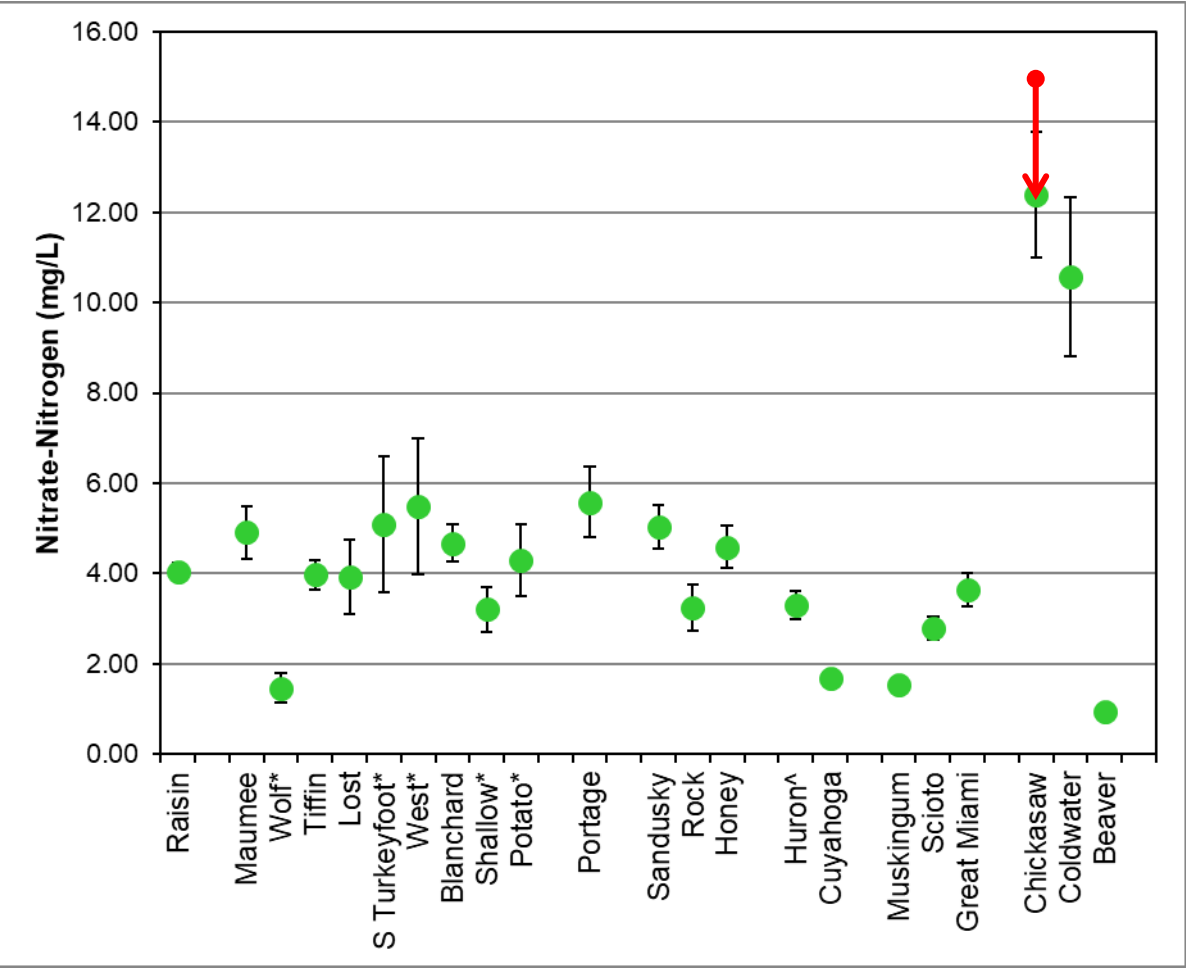
**Grand Lake St Marys inflow tributaries have very high average annual concentrations**



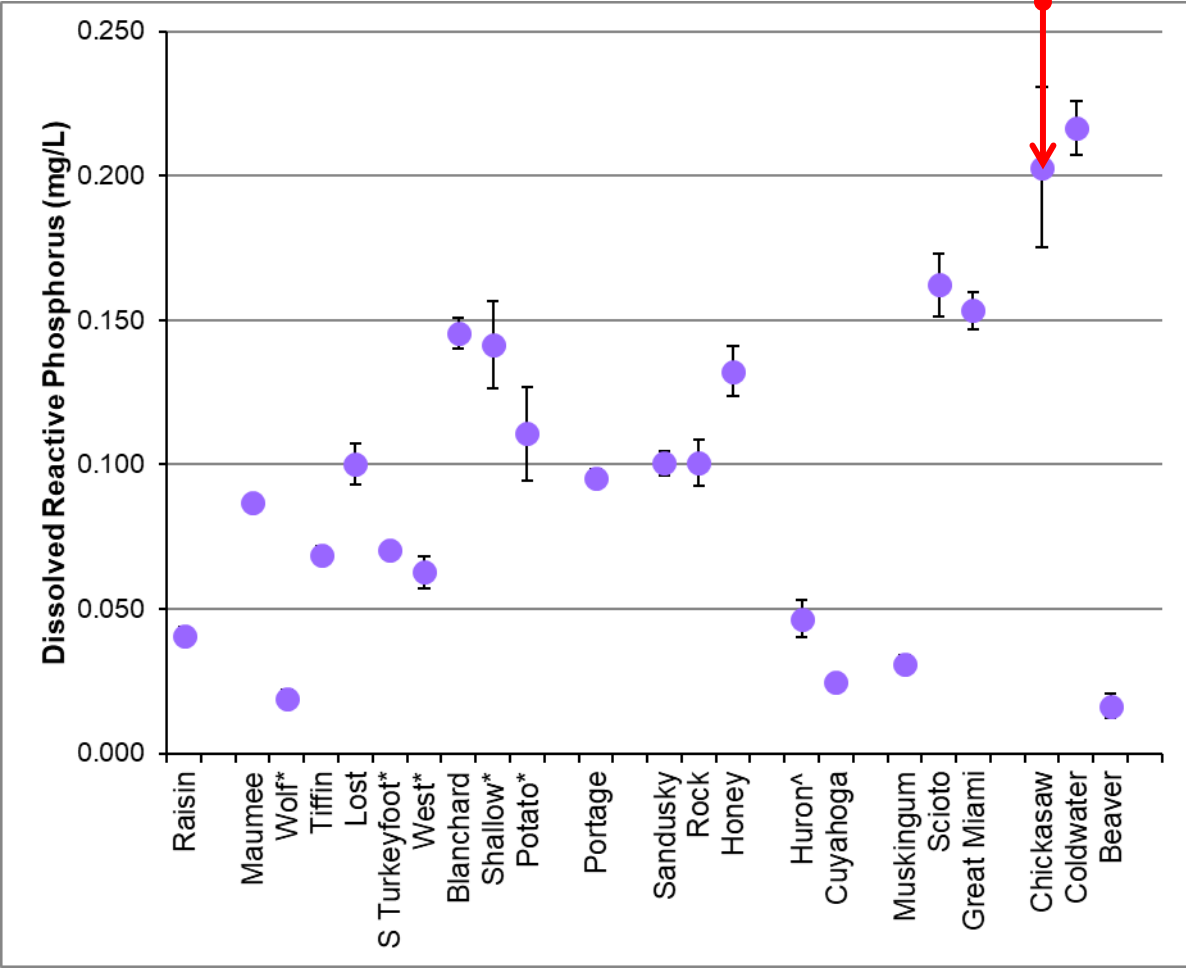
# Change in phosphorus concentrations in Chickasaw Creek due to distressed GLSM watershed designation in 2011



Pre = 2008 - 2011  
Post = 2012 - 2016



| Nitrate-N                           |  |                                 |
|-------------------------------------|--|---------------------------------|
| Average Annual Concentration (mg/L) |  | Percent decrease from 2008-2011 |
| 2008-2011 (pre-manure ban)          |  | 15.4                            |
| 2012-2016                           |  | 10.8                            |
| 2017-2023                           |  | 12.3                            |

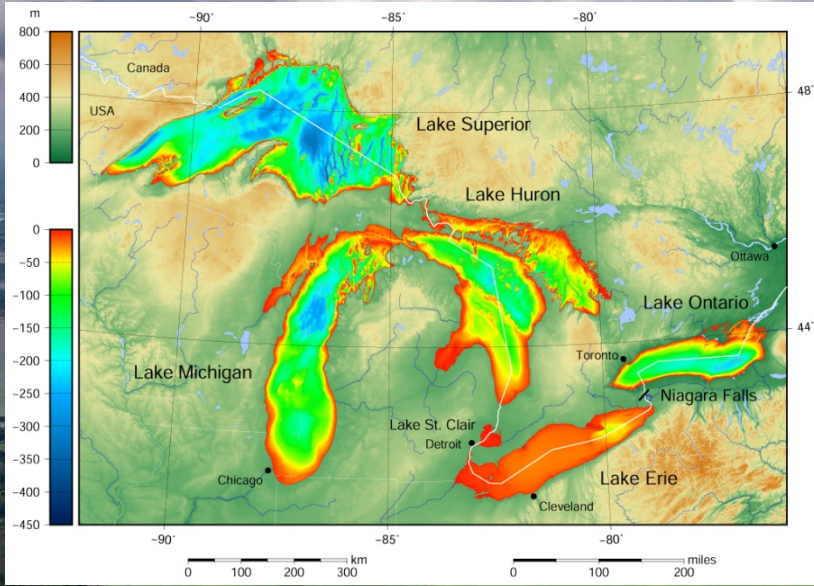


| Dissolved P                         |  |                                 |
|-------------------------------------|--|---------------------------------|
| Average Annual Concentration (mg/L) |  | Percent decrease from 2008-2011 |
| 2008-2011 (pre-manure ban)          |  | 0.265                           |
| 2012-2016                           |  | 0.253                           |
| 2017-2023                           |  | 0.196                           |





**What is causing algal blooms in  
Lake Erie?**



# What are harmful algal blooms?

*\*produce a toxin*  
*\*Microcystis, spp.*

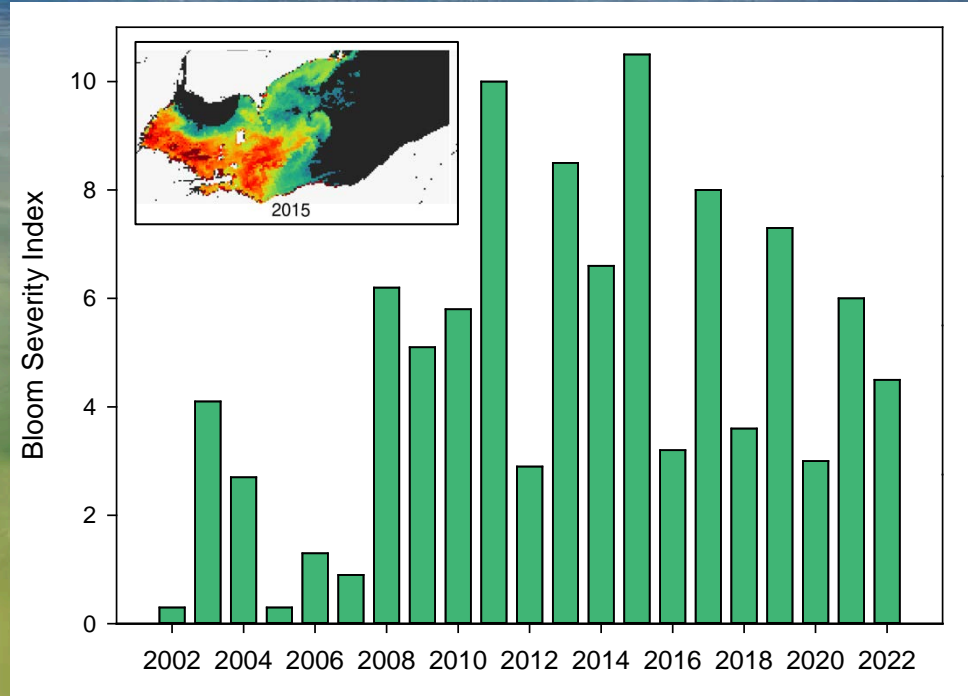
- Light
- Warm temperatures
- Nutrients (nitrogen and phosphorus)

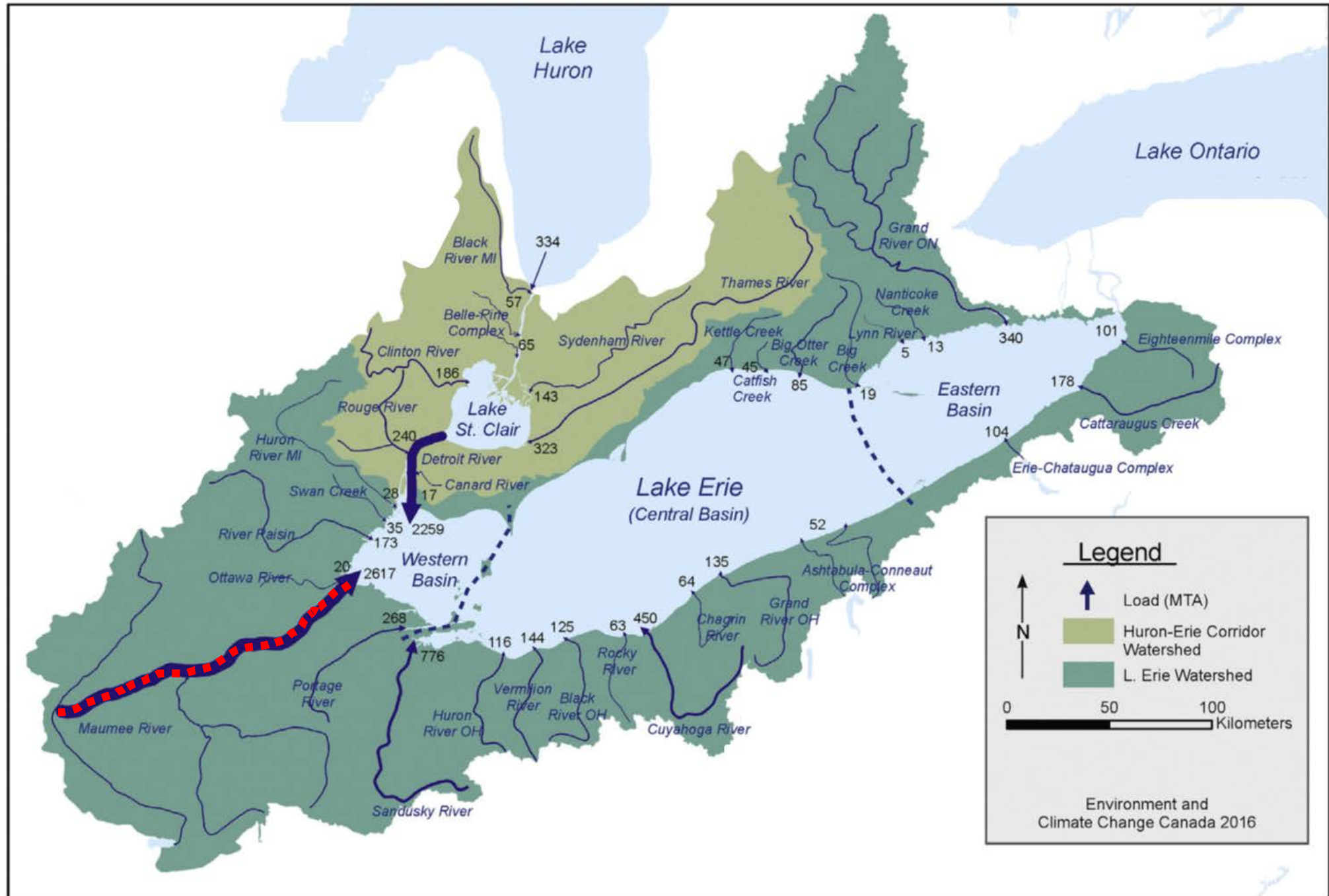


1971



2019

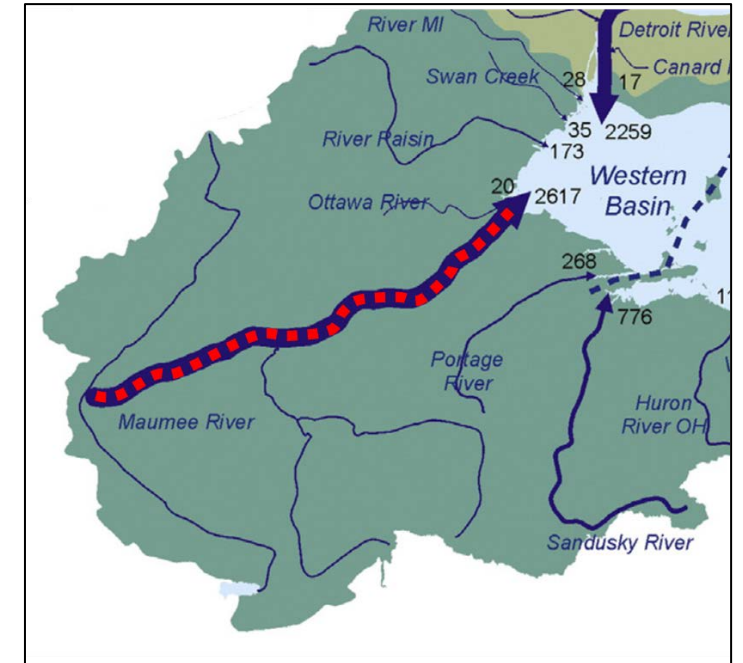
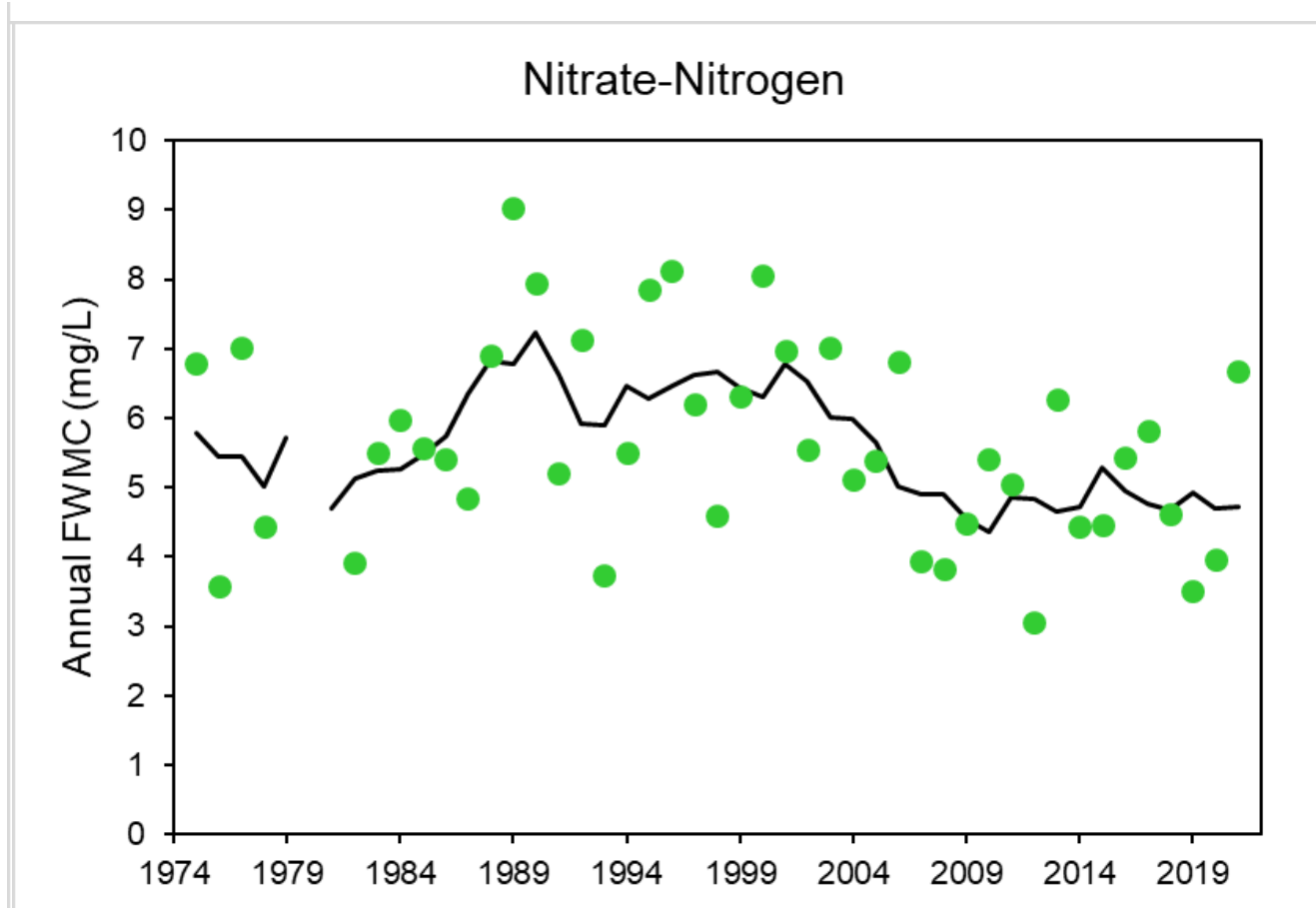




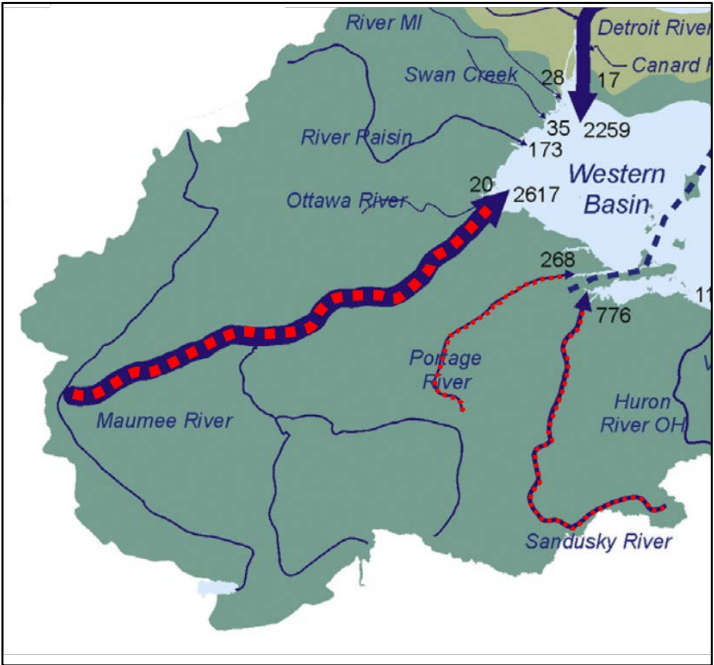
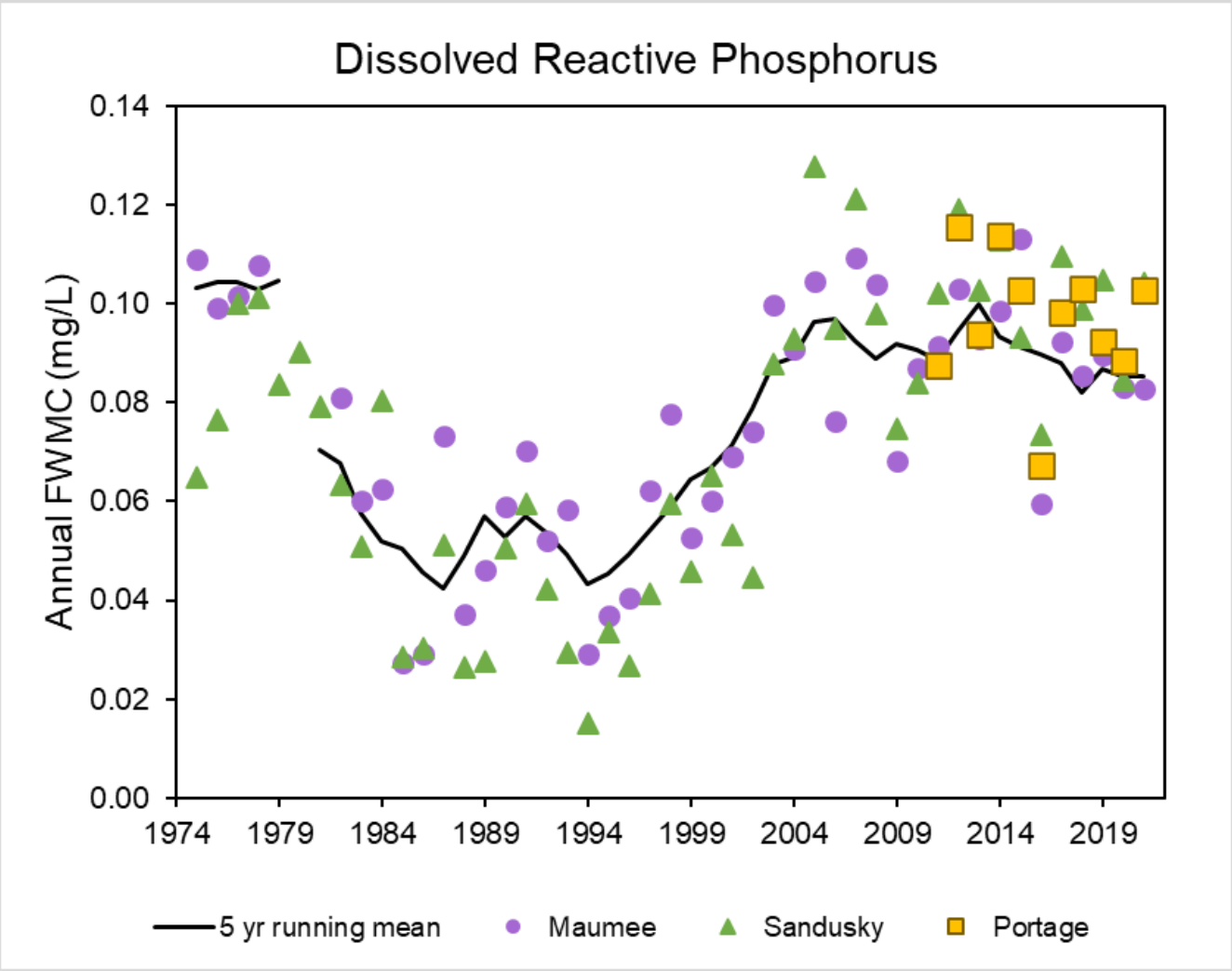
Mean total phosphorus loadings (MTA) to Lake Erie (2003–2013)

Maccoux et al. 2016, JGLR

# Long-term trends in annual nutrient concentrations from the Maumee River

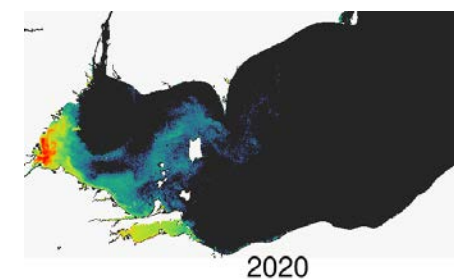
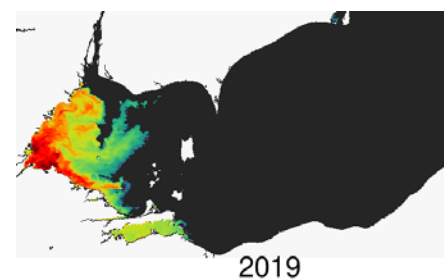
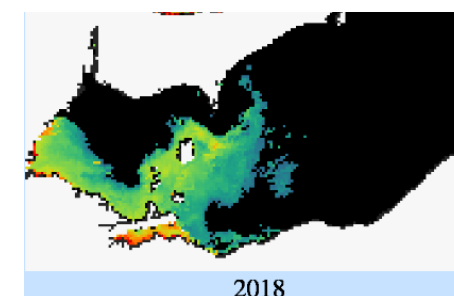
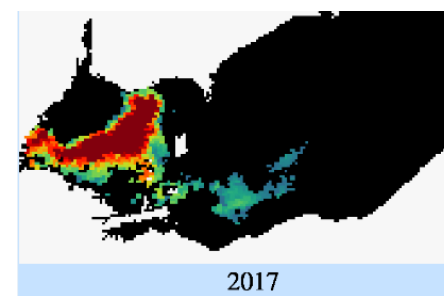
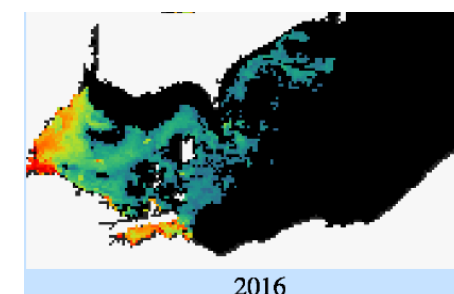
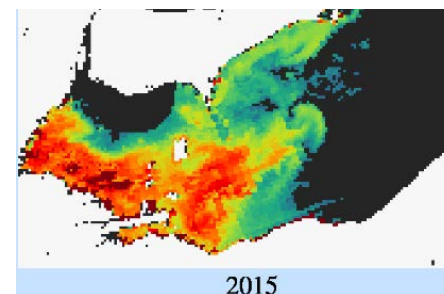
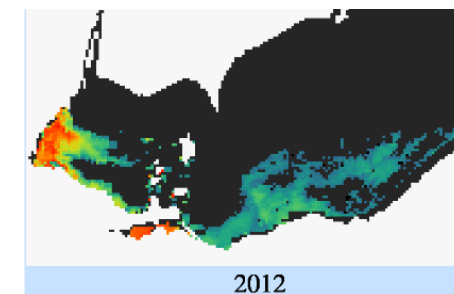
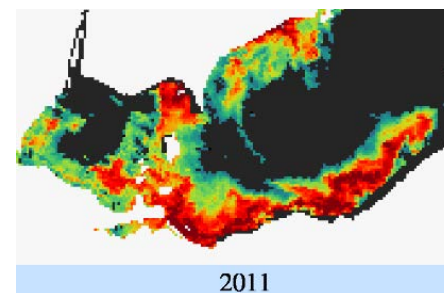
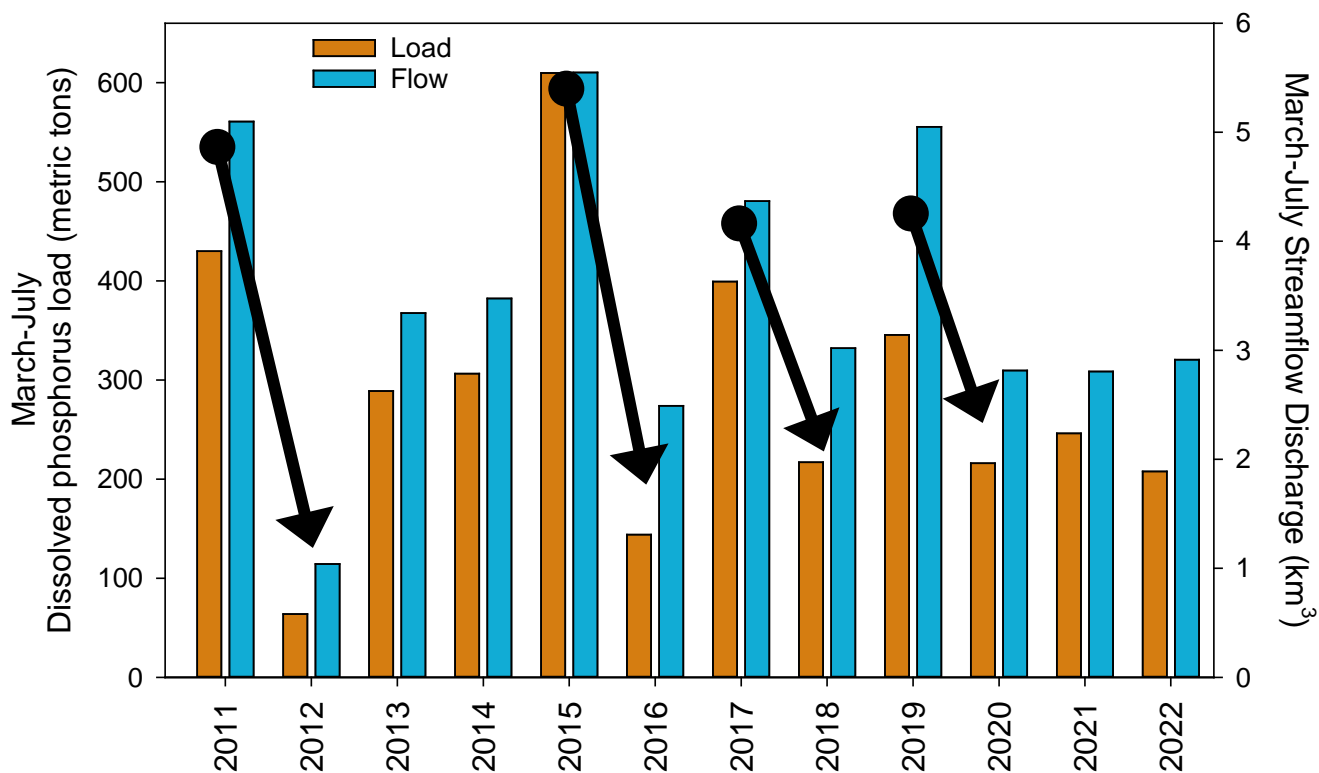


# Similar increases in dissolved phosphorus are found in other WLEB tributaries



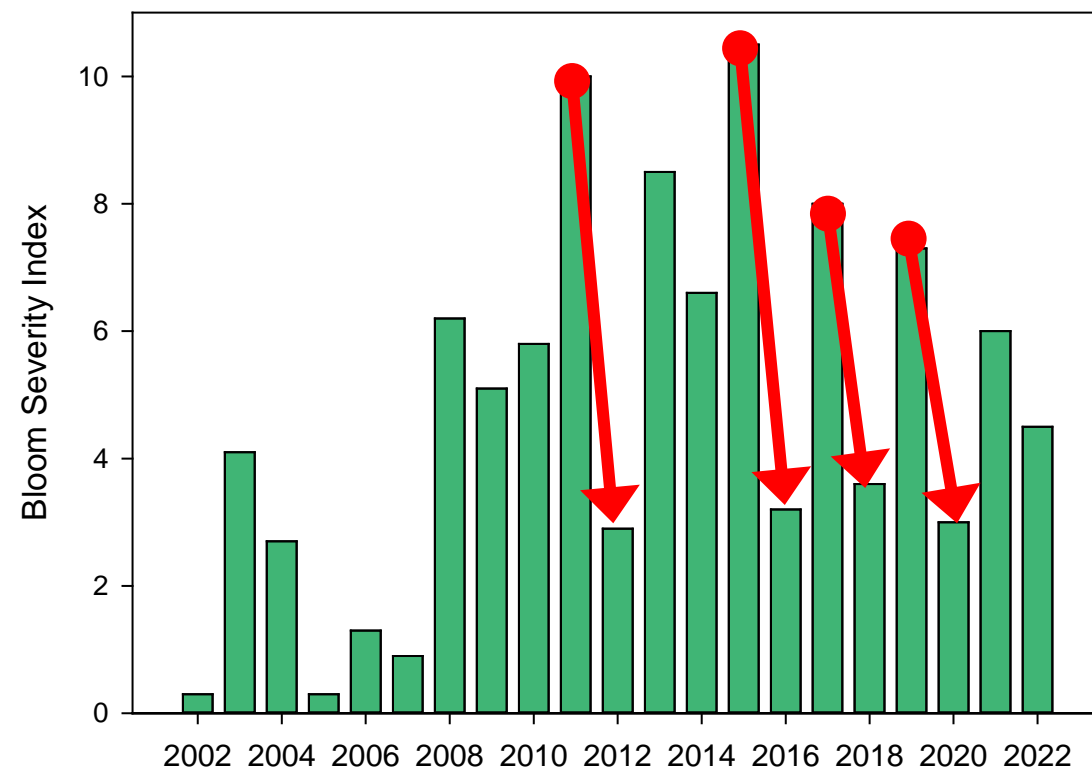
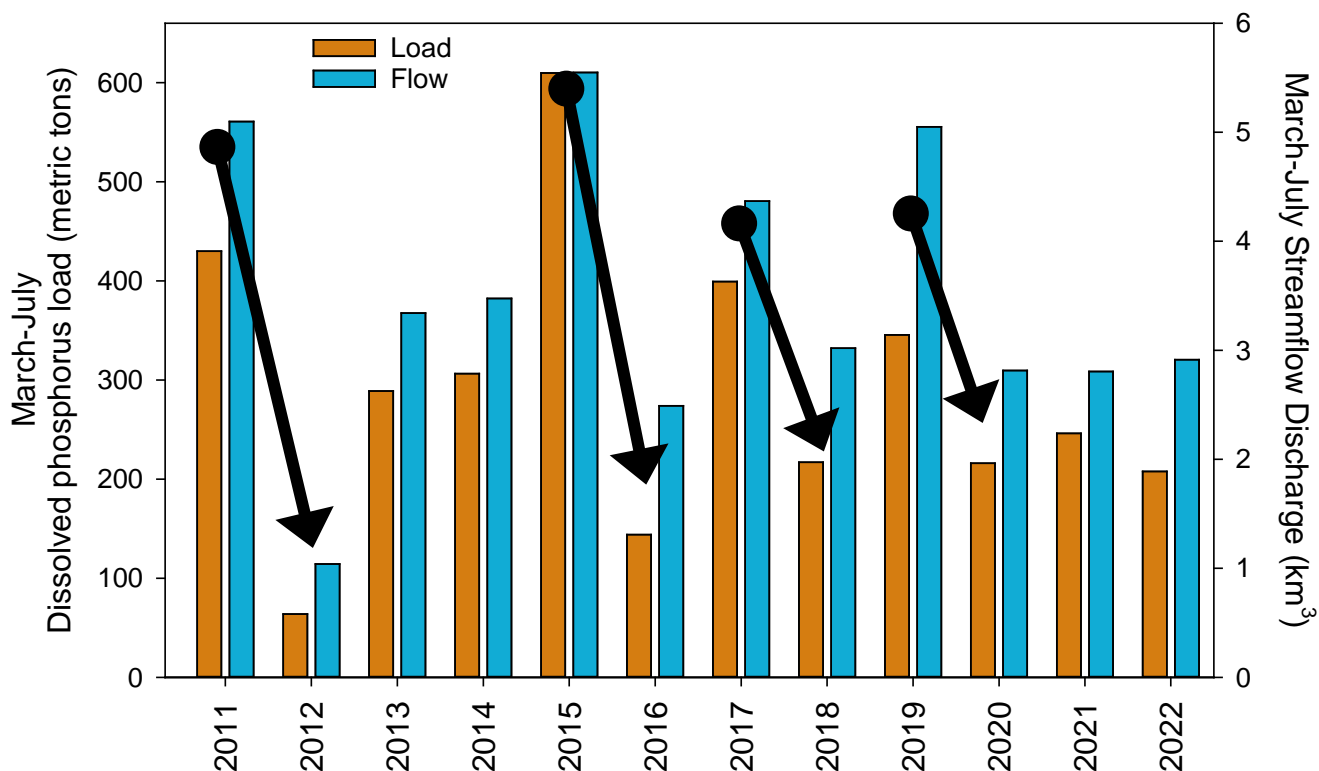
# March to July P loads are linked to bloom severity

Maumee River



# March to July P loads are linked to bloom severity

Maumee River



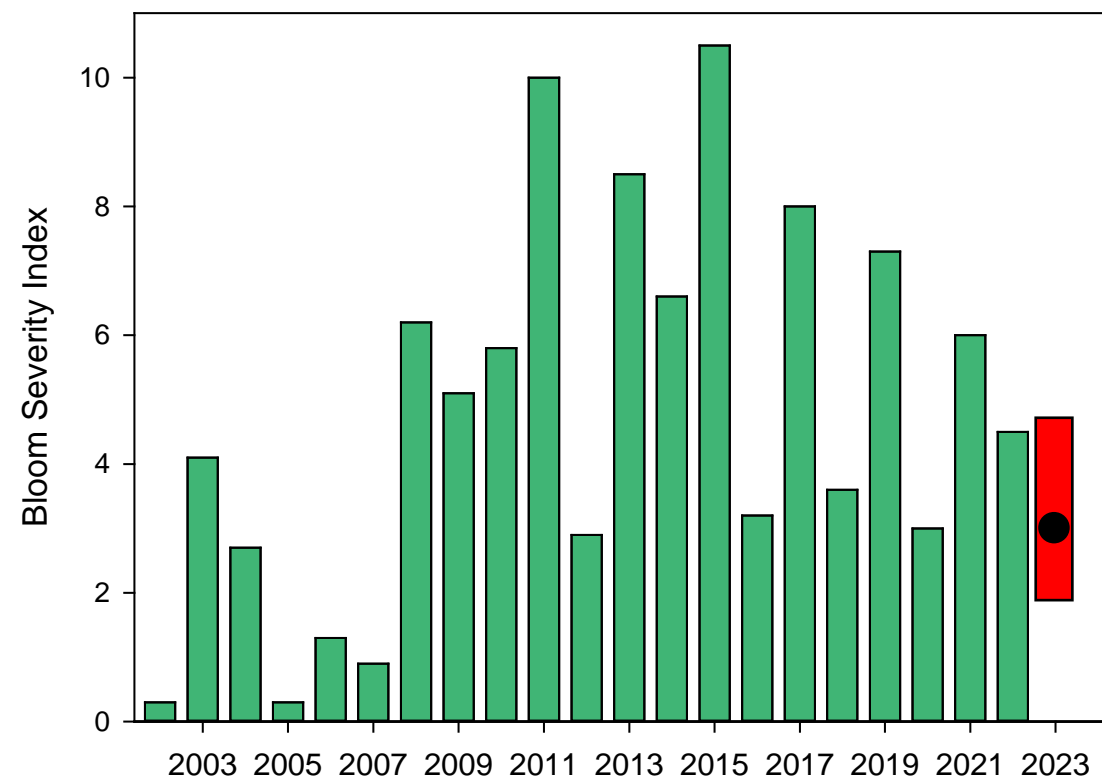
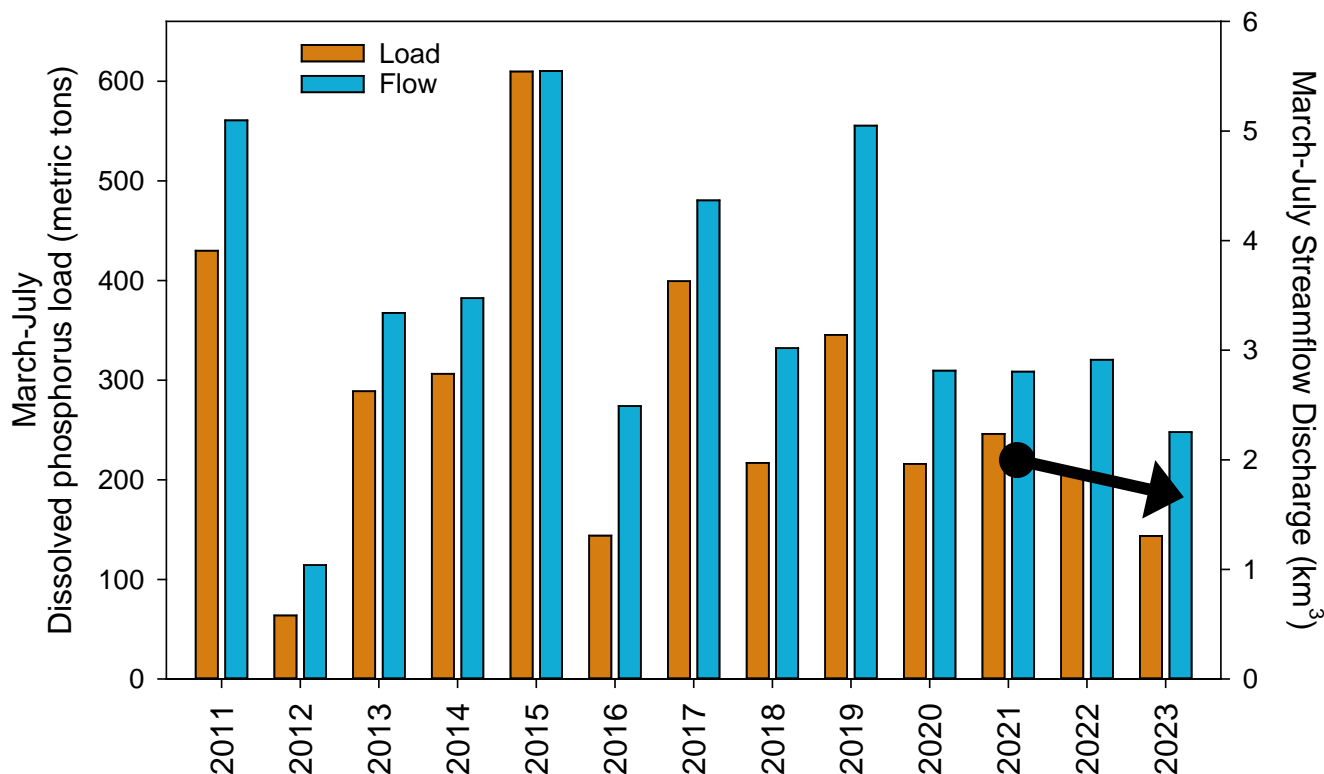
Each summer we use this relationship to forecast bloom severity for Lake Erie

<https://coastalscience.noaa.gov/science-areas/habs/hab-forecasts/lake-erie/>



# The 2023 bloom is expected to be mild (3; 2-4.5)

Maumee River



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<https://coastalscience.noaa.gov/science-areas/habs/hab-forecasts/lake-erie/>



# The Western Lake Erie bloom in 2023 (so far)

June 20

July 4

July 14

July 24

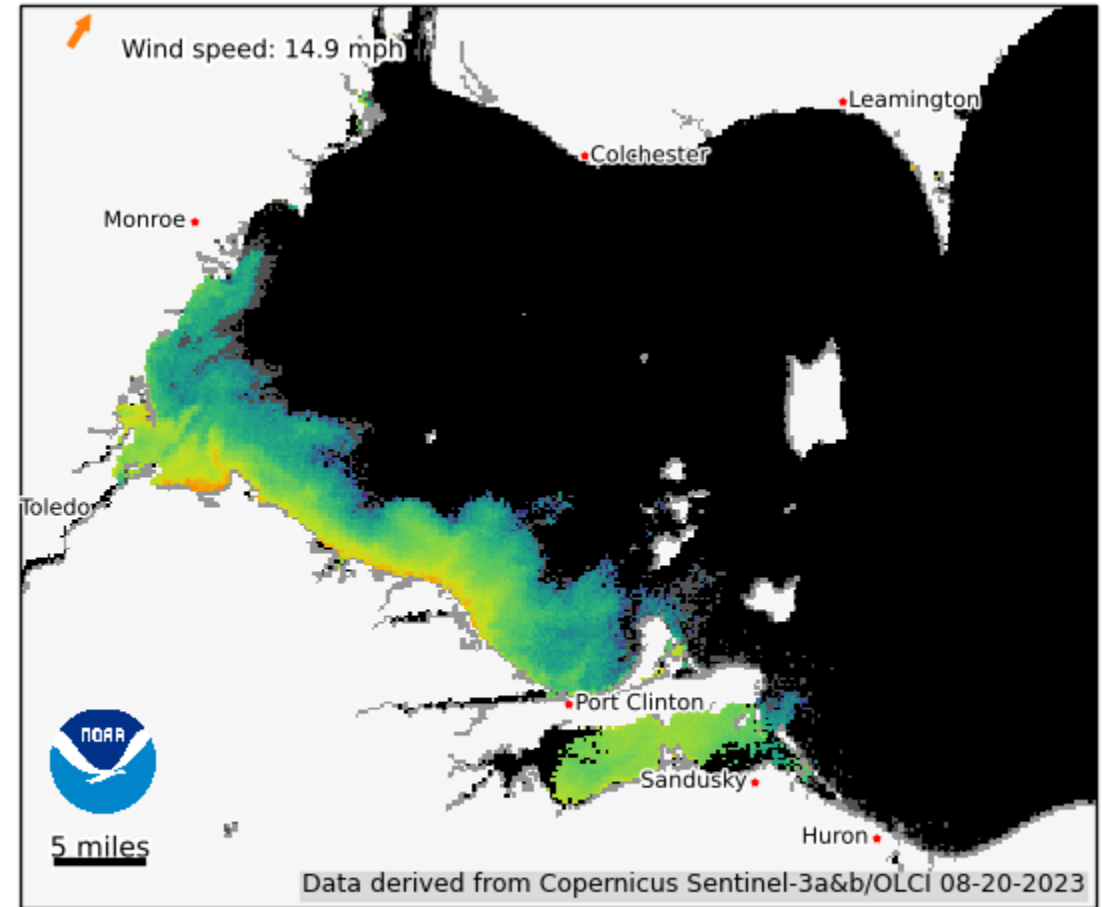
August 4

August 13

August 20



Composited Western Lake Erie basin true color image derived from the OLCI sensor on Copernicus Sentinel-3a&b obtained from EUMETSAT.




Composited Cyanobacteria Index (CI-CIcyano) for Western Lake Erie basin. The algal bloom is present but cloud cover or winds above 9.0 mph prevent determining an area (previous area from Aug 19 was 360 square miles). Winds above 4.0 mph may begin mixing the bloom and clouds may obscure it, leading to an underestimate of the area. Moderate and low concentrations may not be obvious to the eye. Average wind for preceding 3 hours of satellite observation from NOAA NDBC station 46003.



**Why did dissolved  
phosphorus increase?**



The background features a repeating pattern of grey silhouettes of human heads in profile, facing right. Each head contains a large black question mark. The heads are arranged in a staggered, overlapping fashion. The background color is a dark teal or blue-grey. In the center, the text 'What are some solutions?' is written in a bold, white, sans-serif font. A thin white horizontal line is positioned directly beneath the word 'solutions?'.

**What are some  
solutions?**

---

## Point sources



- Reduce nutrients in effluent through technological advances

## Nonpoint sources



- Apply less fertilizer and place it below the surface of the soil
- Reduce water movement off fields

# Reduce nutrient loss



**Soil testing:**  
Testing results give farmers information on where to place fertilizer and fertilizer application rate.



**Variable-rate fertilization:**  
Applying specific fertilizer levels based on the need of each sub-acre to reduce fertilizer application without risk of losing yield.



**Subsurface nutrient application:**  
Applying specific fertilizer below the surface to reduce nutrient loss.



**Manure incorporation:**  
Mixing manure into the soil to keep it in place and minimize nutrient loss.



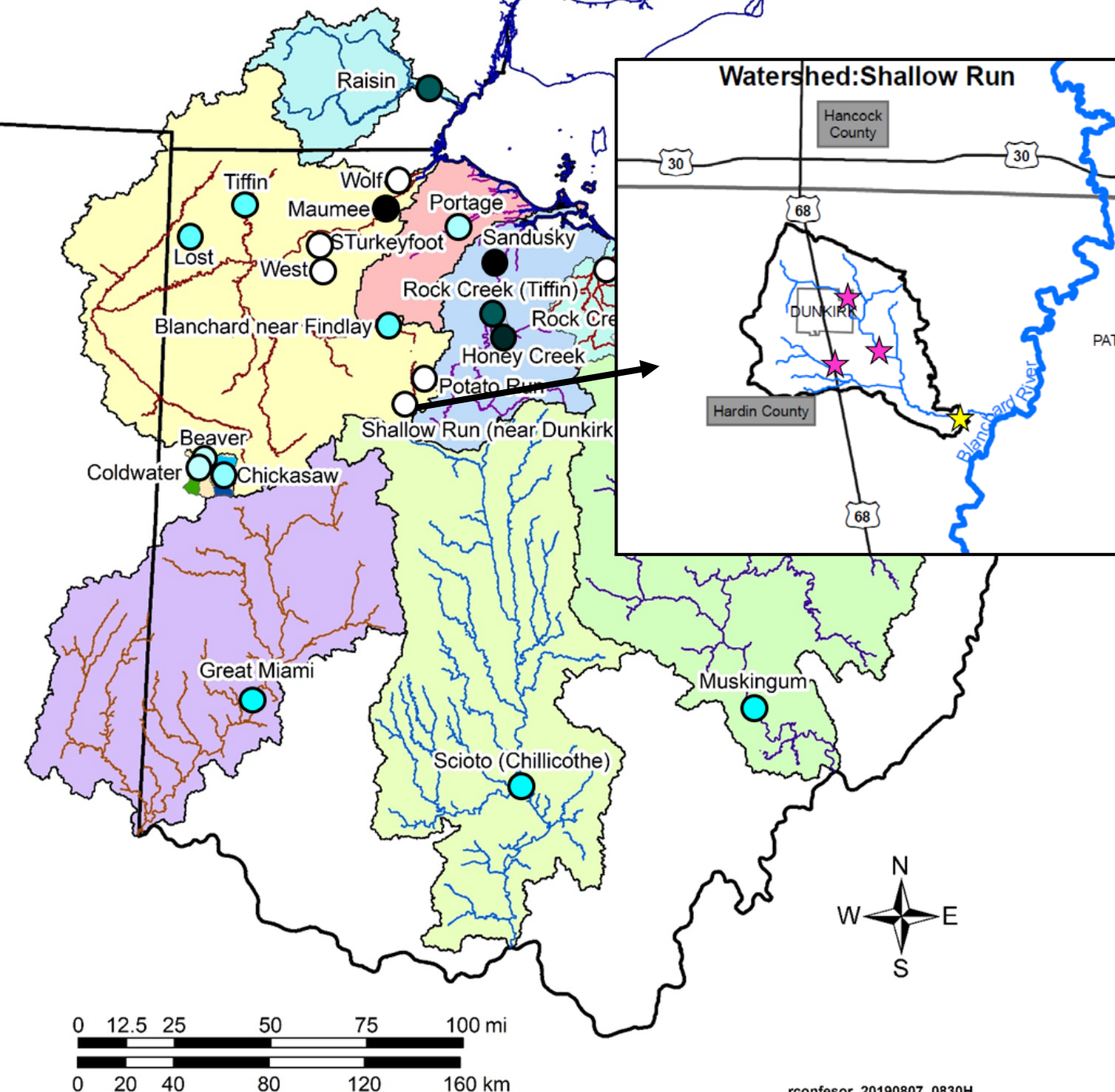
THE OHIO STATE  
UNIVERSITY

# The Pilot Watershed Approach

*Reduce Scale & Increase Adoption*



- Heavily invest in practices, monitoring, outreach in small sub-watershed
- GOAL: 70%+ adoption rates in Shallow Run
- Demonstrate ability of practices to increase water quality: “Can we move the needle?”
- With results, extrapolate to determine successful approaches for larger areas.
- Collaboration critical



## Paired approach

- Shallow Run (5,500 acres) is the treatment
- Potato Run (11,000 acres) is the control
- Monitoring began in 2018
- Practices will focus on reducing dissolved phosphorus
- Implementation will begin this fall



# Reduce erosion and slow down water

5



## **Conservation crop rotation:**

Planting certain crops that reduce erosion and enrich the soil thus reducing runoff and sediment delivery.

6



## **Cover crops:**

When planted after the main harvest, cover crops reduce erosion, hold nutrients in the soil, and improve soil health.

7



## **Drainage water management:**

Slowing down runoff to give phosphorus more time to settle back in the soil.

8



## **Two-stage ditch construction:**

Creating modified drainage ditches to slow water flow and allow the phosphorus to settle.

9



## **Edge-of-field buffers:**

When trees, shrubs or strips of grass are planted along farm fields in the right place, the plants hold on to phosphorus and prevent its release into the water.

10



## **Wetlands:**

Wetland vegetation and soils absorb phosphorus, slow down the movement of water, offer a natural filtering process, and allow phosphorus to settle.

# The H2Ohio Wetland Monitoring Program

Managed by the Lake Erie and Aquatic Research Network (LEARN)  
and the Ohio Department of Natural Resources (ODNR)



## GOAL OF THE PROGRAM

The ultimate goal of the H2Ohio Wetland Monitoring Program is to assess nutrient removal of wetland restoration sites to help improve future restoration and management strategies.

## CORE QUESTIONS

The core questions of the program are:

- ▶ Which types of wetland structure and function provide enhanced nutrient reduction and retention?
- ▶ Which wetland restoration approaches maximize cost-effectiveness to mitigate nutrient loads to Ohio water bodies?
- ▶ How can wetland restoration be effectively implemented in the future?

## Program Partners

Lake Erie and Aquatic Research Network  
Ohio Department of  
Natural Resources  
The Ohio State University  
College of Food, Agricultural,  
and Environmental Sciences

Ohio Sea Grant and Stone Lab  
Kent State University  
The University of Toledo  
Heidelberg University National  
Center for Water Quality Research

Old Woman Creek National  
Estuarine Research Reserve  
Bowling Green State University  
Wright State University

### and Water Quality Incentive Program Projects



Sandusky  
Headwaters  
Preserve

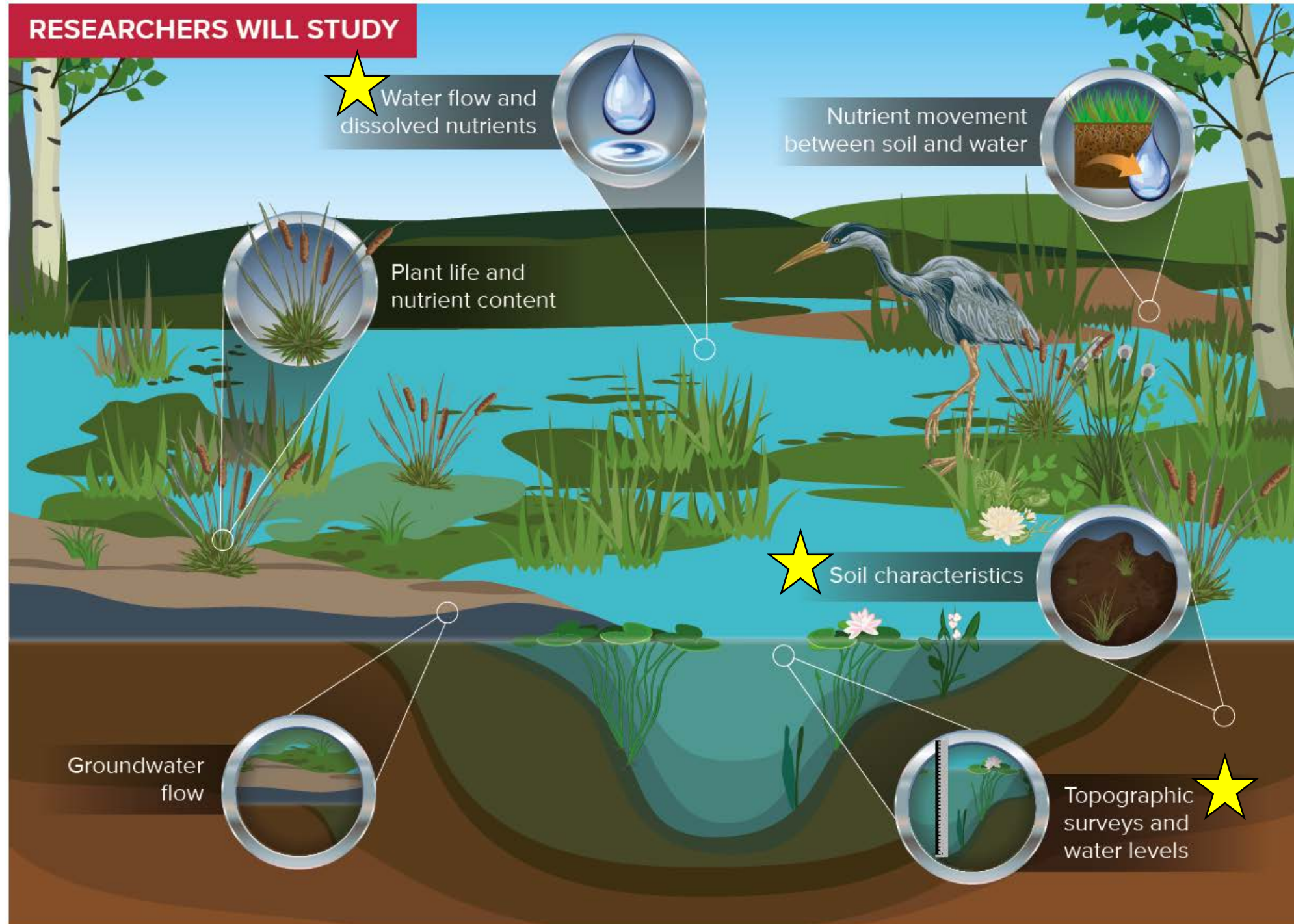
# Buehler Farms

## Redhorse Bend

## Fruth Wetland



# RESEARCHERS WILL STUDY



## YEAR 1 MONITORING EFFORT

More than 1300 water samples from 34 wetlands



Studies of nutrient movement between soil and water



Plant life surveyed in 11 wetlands, more than 800 plant samples collected



More than 33 technicians and students hired and trained

More than 500 soil samples from 34 wetlands

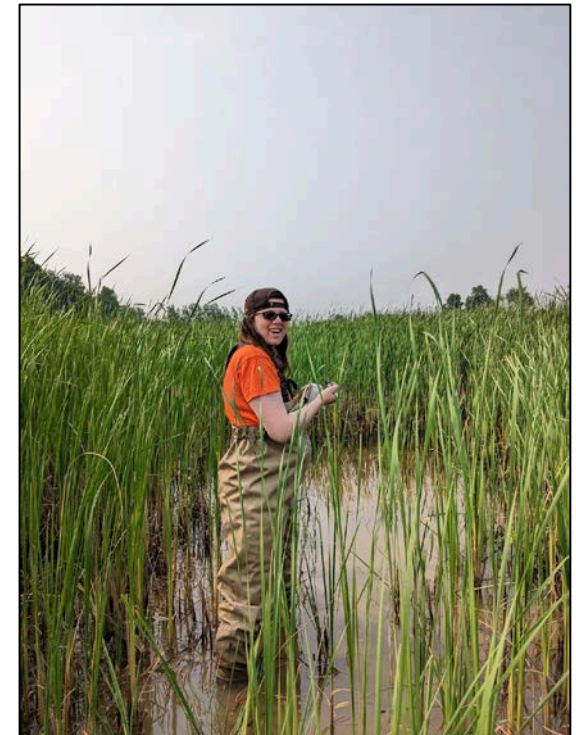
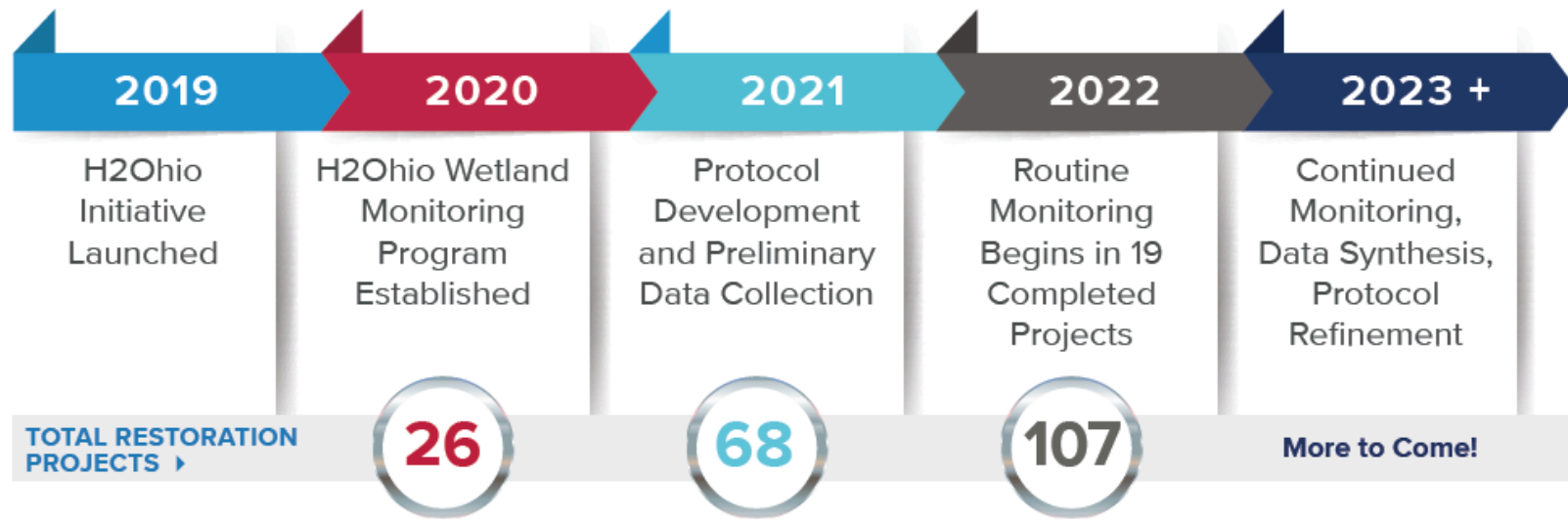


Overview of groundwater flow



Production of topographic surveys and water levels





# Take Home

- To best understand the threats and solutions to water quality issues, a rigorous water quality monitoring program is necessary
- Threats to water quality come from many different sources and vary by watershed
- Improving water quality associated with nonpoint sources can be challenging



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<http://www.ncwqr.org>  
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<https://seagull.glos.org/>